



# MICROMACHINING TECHNOLOGY

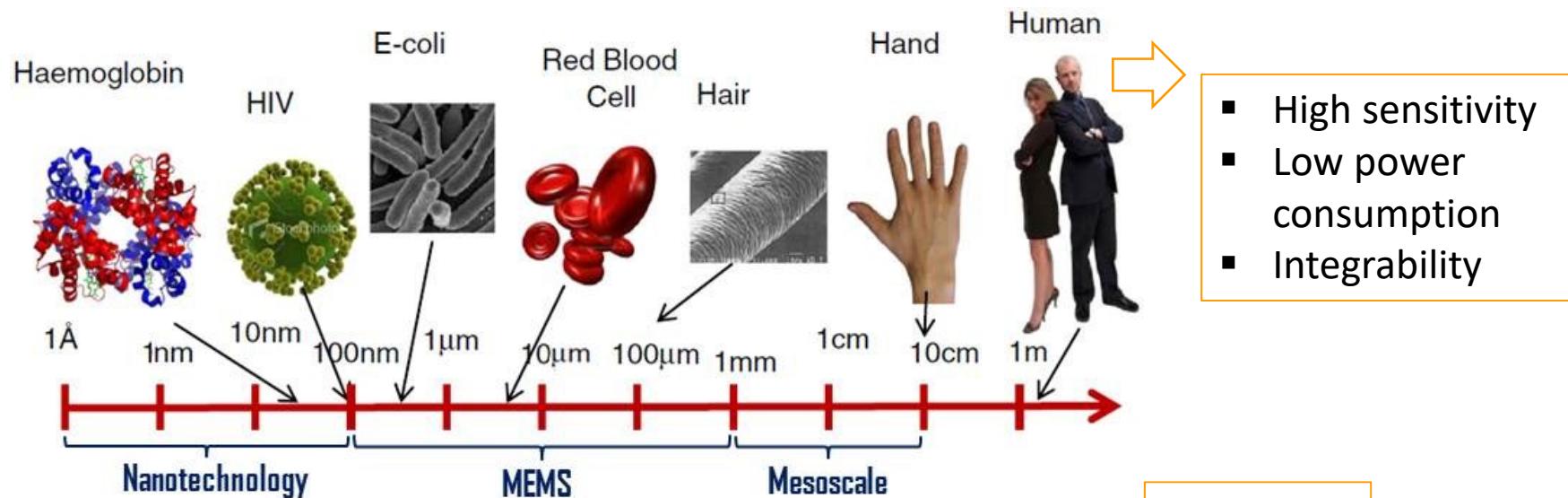
Péter Fürjes

E-mail: furjes@mfa.kfki.hu



# MEMS: Micro-ElectroMechanical Systems

**Miniaturised devices and systems:** in the range between 100nm and 1000μm



## Fabrication technology: SILICON micromachining

- Photolithography
  - Physical and chemical layer deposition (metals, dielectrics)
  - Wet and dry etching

*SolidState Technology, Ramesh Ramadoss, MEMS devices for biomedical applications*  
<http://electroiq.com/blog/2013/10/mems-devices-for-biomedical-applications/>



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# MICROMECHANICS



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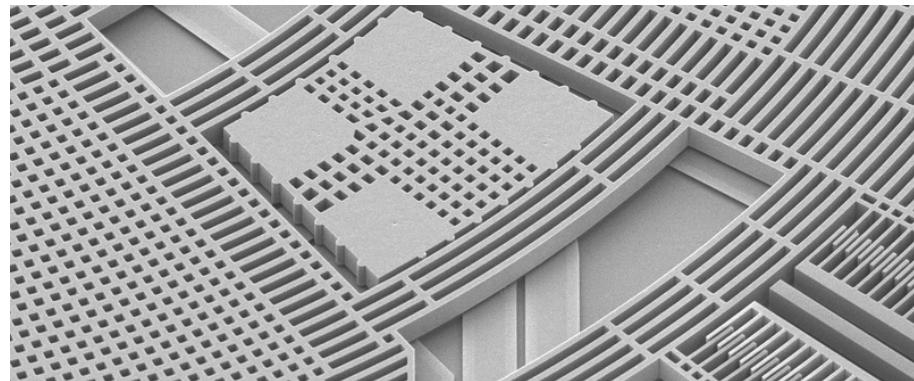
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MEMS: „2D” IC technology  3D structures

- membranes, suspended structures, movable elements,
- microfluidic applications: channels, chambers, reactors etc.

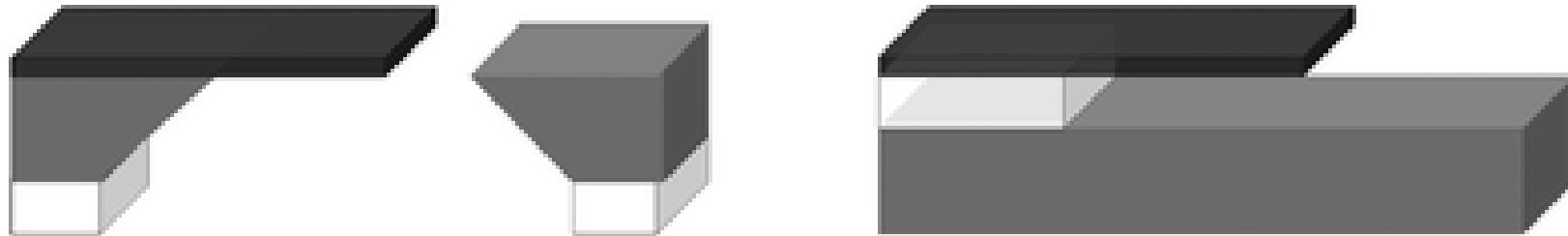


## Microfabrications:

- processes and devices: different from traditional mechanical fabrication technologies
- mainly „dry” and „wet” chemical etching and electrochemical methods,  
BUT classic processes (laser or diamond blade cutting)

Typical dimensions: 1-500 mm

Thickness of the Si crystal: 380-500-1000 mm

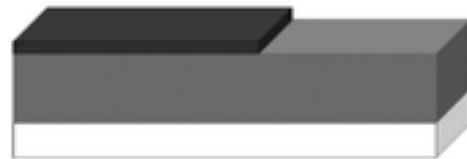


	Bulk	Surface
<i>Dimensions</i>	$2-3 \mu\text{m} < a < 100-500 \mu\text{m}$	$a < 2-3 \mu\text{m}$
<i>Thermal isolation</i>	+	-
<i>Mechanical stability</i>	+	-
<i>Membranes</i>	Single crystalline	amorphous or polycrystalline

{ **3rd solution:** Thin single crystalline layers: "Smart Cut" / SOI (silicon-on-insulator) }



- Device Layer
- Silicon Wafer
- Silicon Oxide

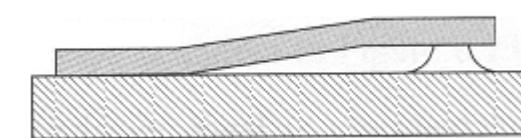
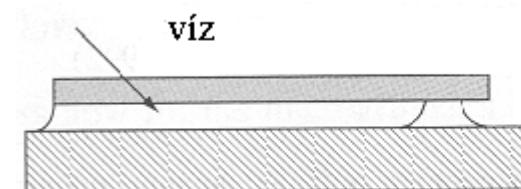


(a)

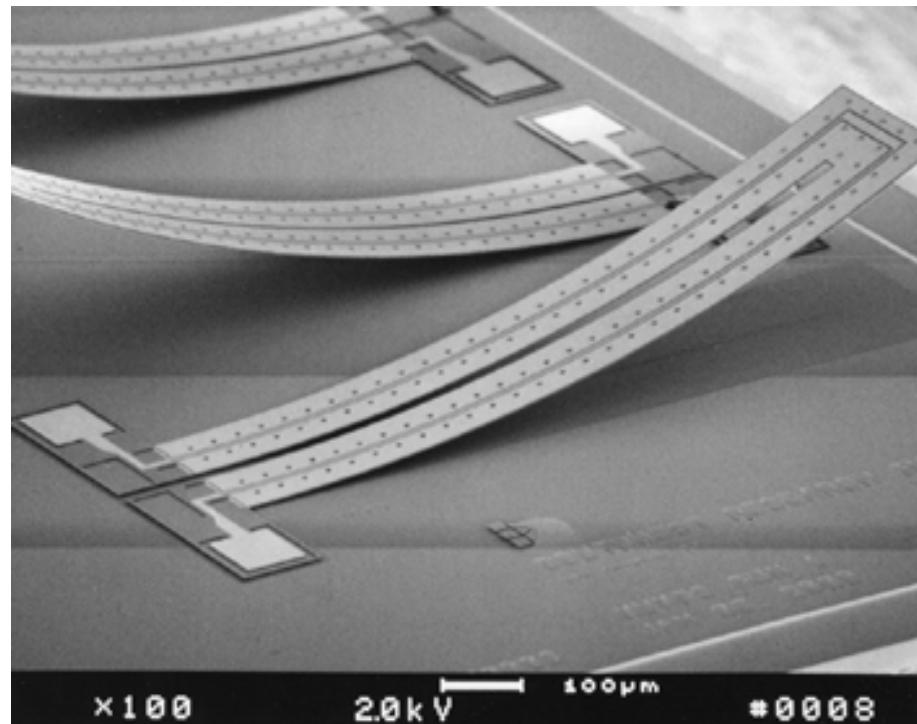
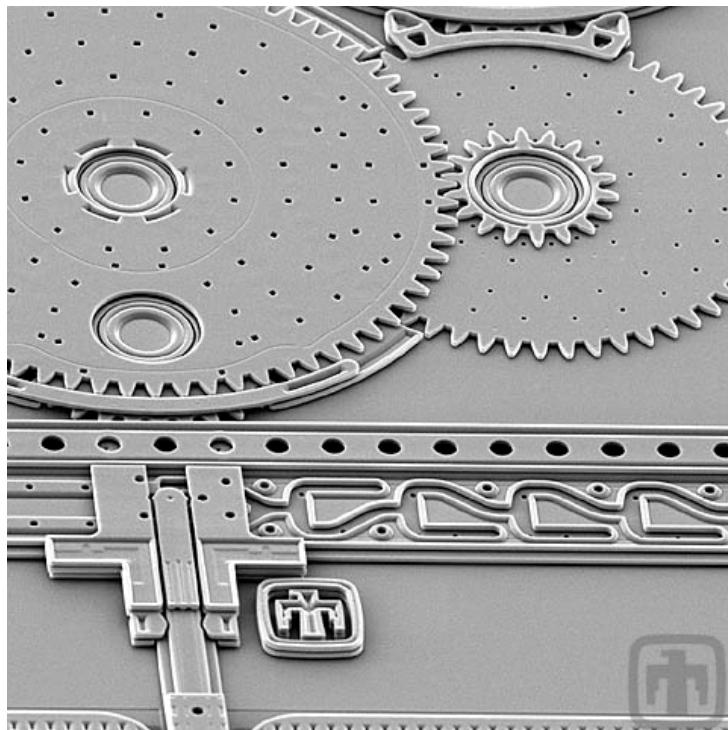


(b)

Typical problem: sticking



Solution:  
inbuilt keeper  
or perforated structures  
or dry etching  
or supercritical drying

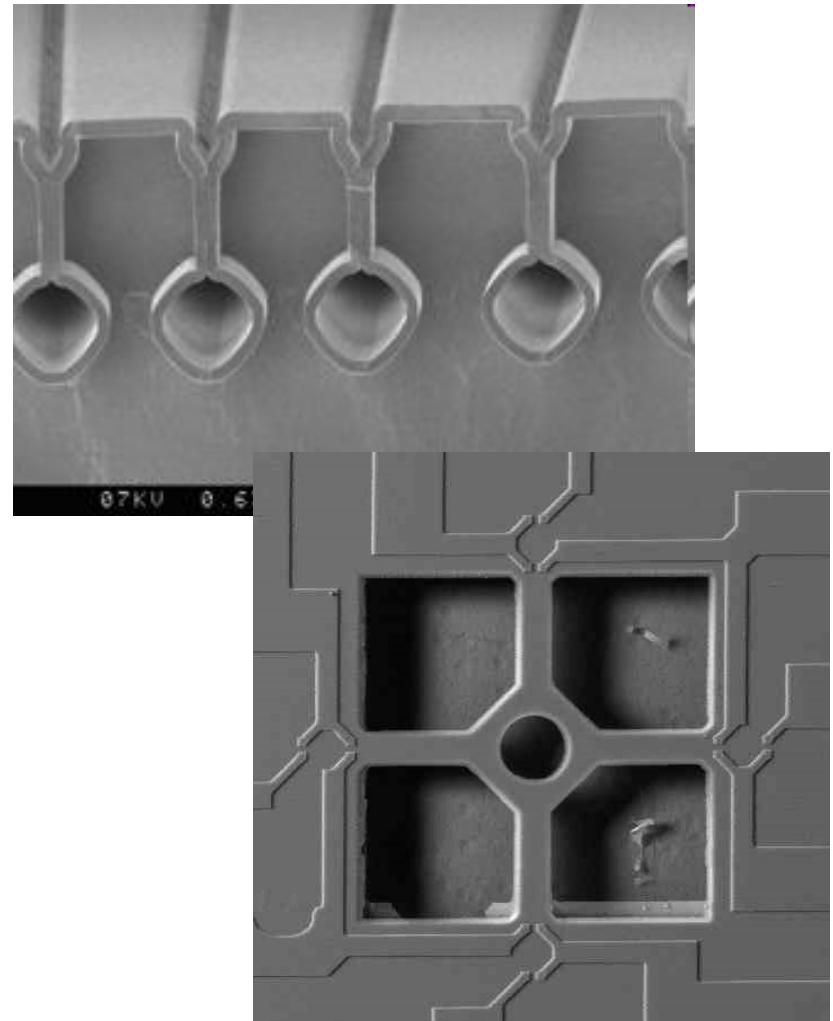
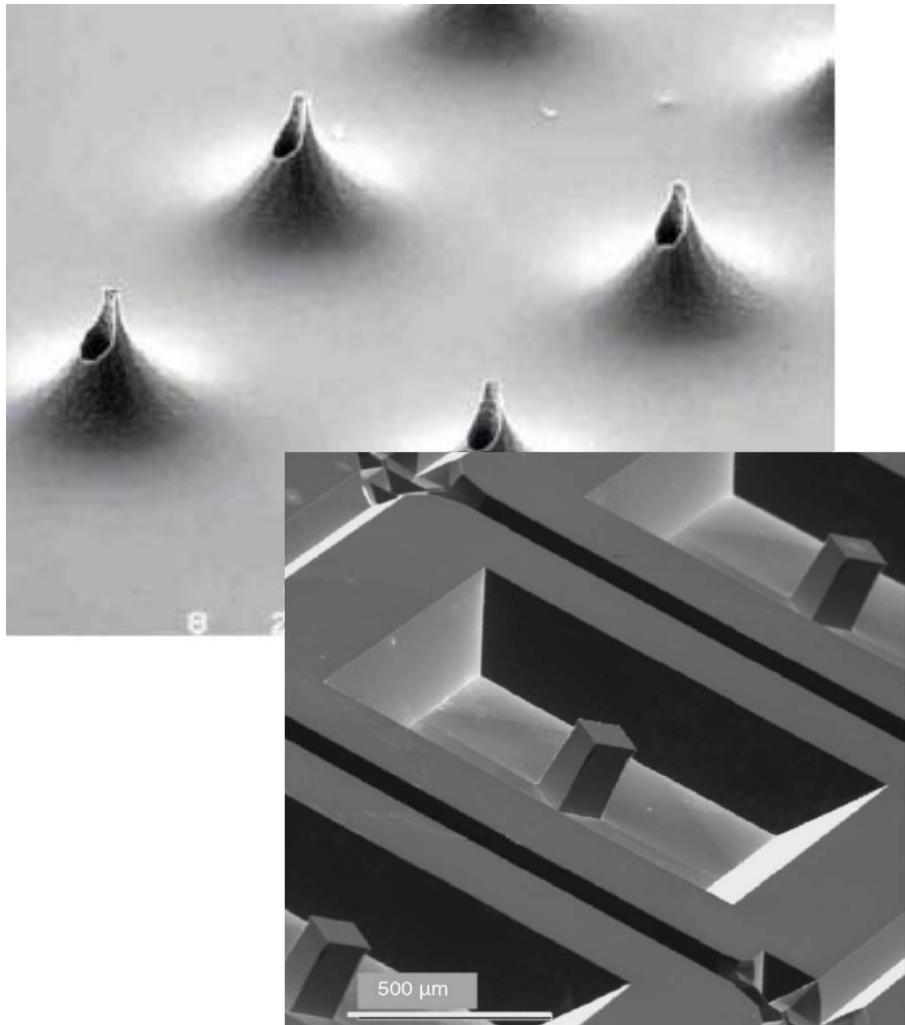




## EXAMPLES: BULK MICROMACHINING

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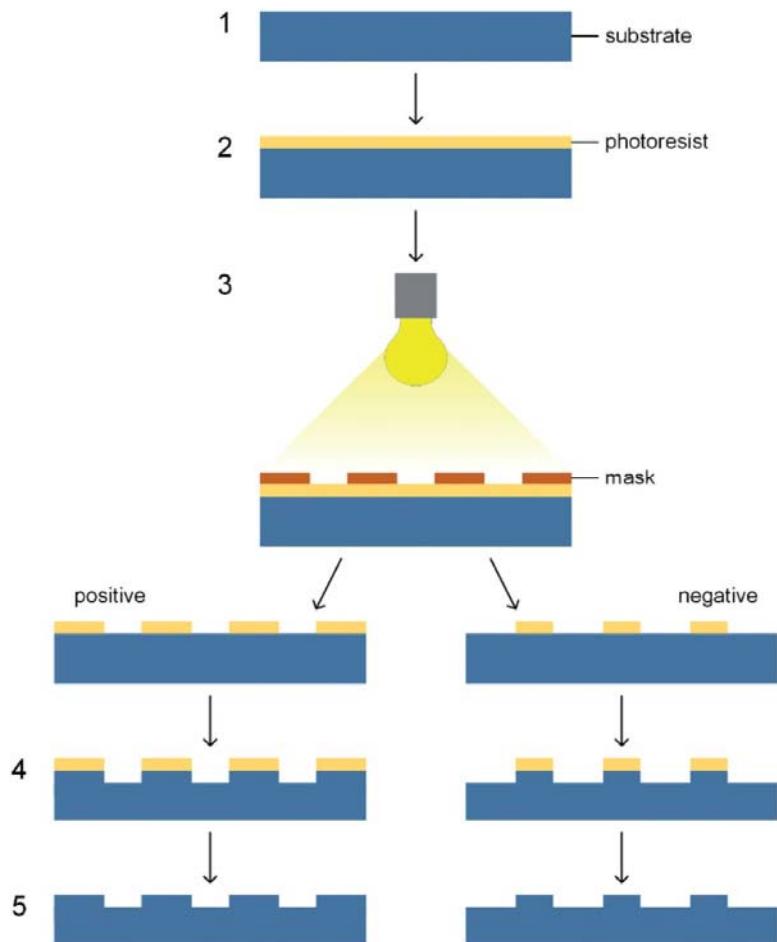
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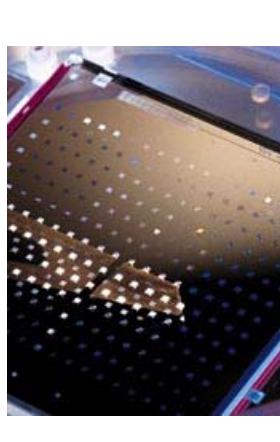
# PHOTOLITHOGRAPHY

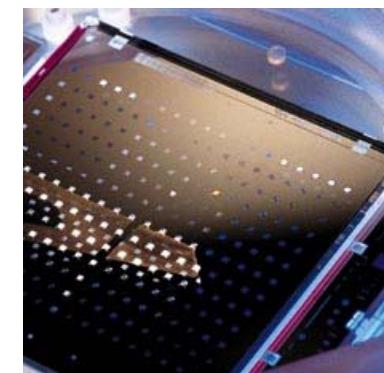
# SUBSTRACTIVE PHOTOLITHOGRAPHY



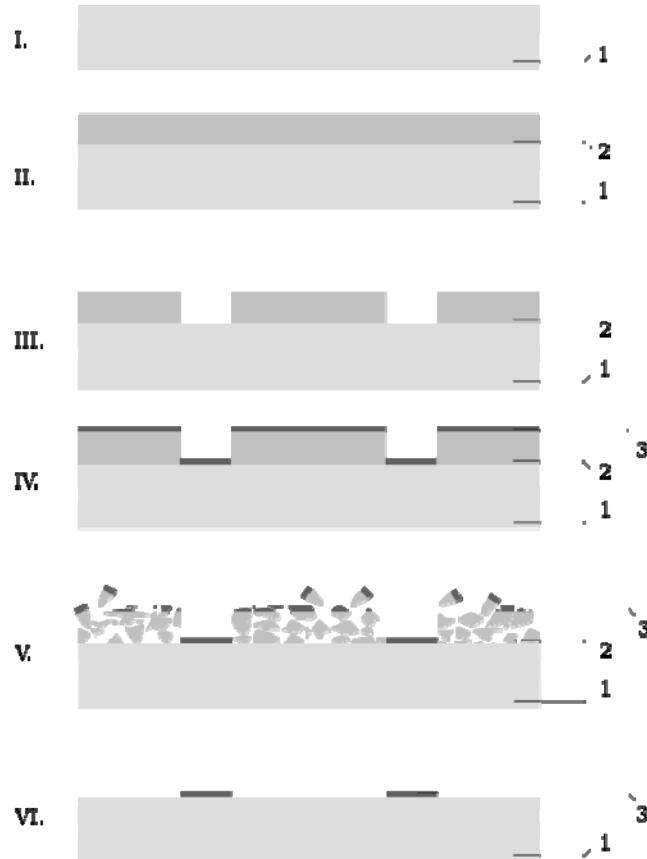
1. Surface treatments: cleaning, dehydratation
  2. Photoresist spincoat / prebake
  3. Exposure / development

Postexposure  
bake / softbake  
hardbake

  4. Processing with photoresist masking
  5. Photoresist removal, stripping, cleaning



# ADDITIVE PHOTOLITHOGRAPHY

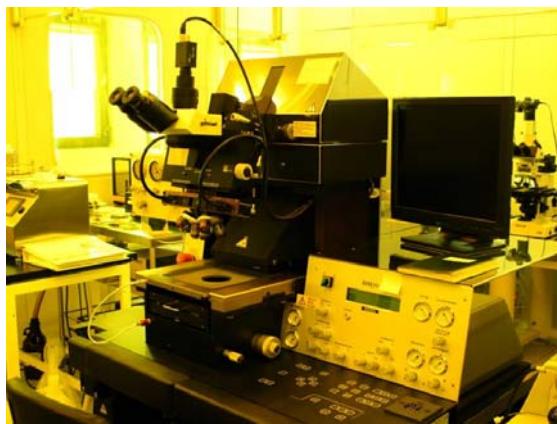


1. Surface treatments: cleaning, dehydratation
  2. Photoresist spincoat / prebake
  3. Exposure / development  
Postexposure  
bake / softbake
  4. Layer deposition
  5. Photoresist removal, stripping,  
cleaning

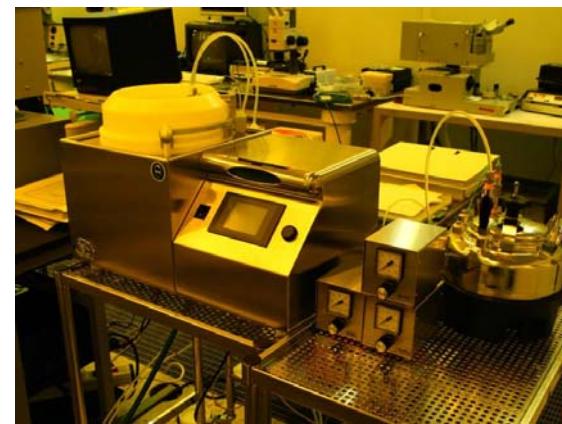
# EQUIPMENTS - RADIATION



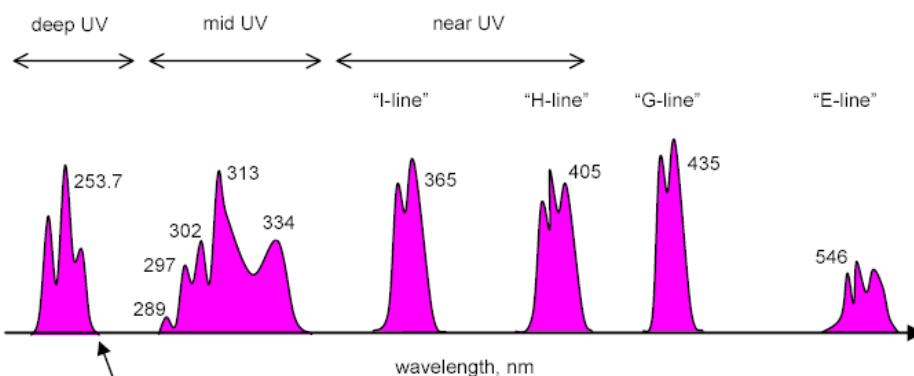
## Spincoater – hotplate



## mask aligner



## developer



Hg lamp: 436 nm (g-line), 405 nm (h-line), 365 nm (i-line)

KrF laser: 248 nm / ArF lézer: 193 nm

Next generation: extreme UV (EUV): 13.5 nm

# LAYER DEPOSITION



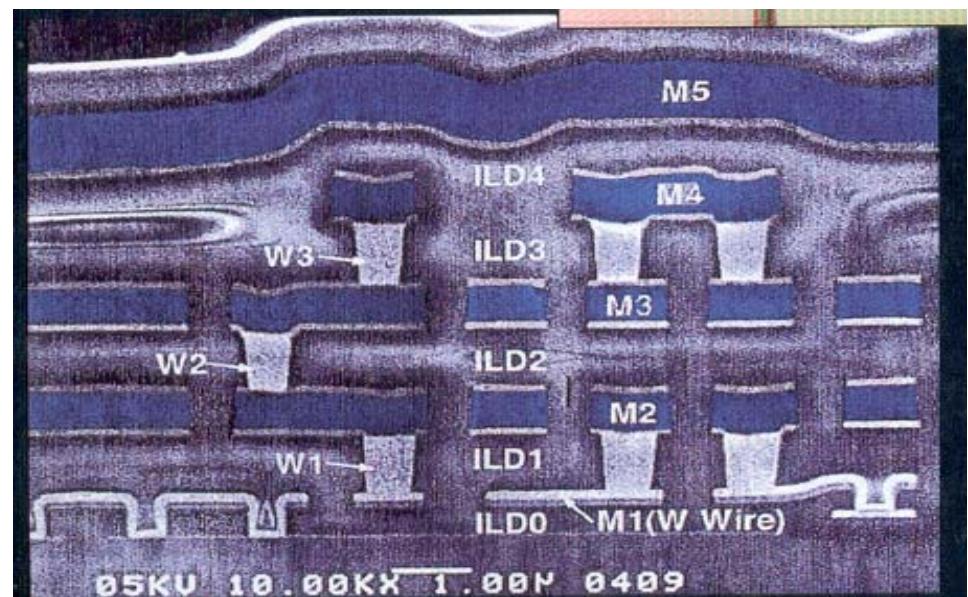
## APPLICATIONS

- Microelectronics, semiconductor processing
- Micro-electromechanical systems (sensors, actuators, MEMS)
- Thermal conducting coatings (BeO, AlN, diamond)
- Photovoltaic devices (solar cells)
  - amorphous and microcrystalline Si layers on glass and polymer substrates
  - compound-semiconductors (CuInGaSe, CdTe)
  - single- and multicrystalline Si solar cells (HIT)
- Optical applications (filters, gratings, antireflexion layers, mirrors, etc.)
- Abrasion-resistant coatings
  - protection of optical devices (deposited diamond layers)
  - hard coating of tools (TiN, WC, B<sub>4</sub>C, diamond, DLC)
  - coatings of human prosthesis
- Corrosion-resistant coatings
- Decoration coatings



# STANDARD REQUIREMENTS

- homogeneous thickness on the substrate
  - homogeneous composition
  - homogeneous structure (amorphous, polycrystalline, epitaxial)
  - homogeneous physical and chemical properties
  - compactness (sponge vs. layer, pinholes)
  - adequate adhesion
  - low thermomechanical stress
  - special requirements  
(friction, wettability,  
biocompatibility, etc..)
  - economical
    - deposition rate
    - infrastructural maintenance
  - step coverage





## TECHNOLOGIES

### Physical methods (PVD, Physical Vapour Deposition)

Solid source:

vacuum evaporation  
sputtering: DC, RF, magnetron  
MBE (Molecular Beam Epitaxy)  
LPE (Liquid Phase Epitaxy)  
(single crystal growing, Czohralsky, Floating zone)

Melt source:

### Chemical methods

Electrolite source:

plating  
(solution, suspension)

gázfázisból:

CVD (Chemical Vapour Deposition)

VPE (Vapour Phase Epitaxy)

MOCVD (Metal Organic ....)

LPCVD (Low pressure...)

PECVD (Plasma enhanced...)

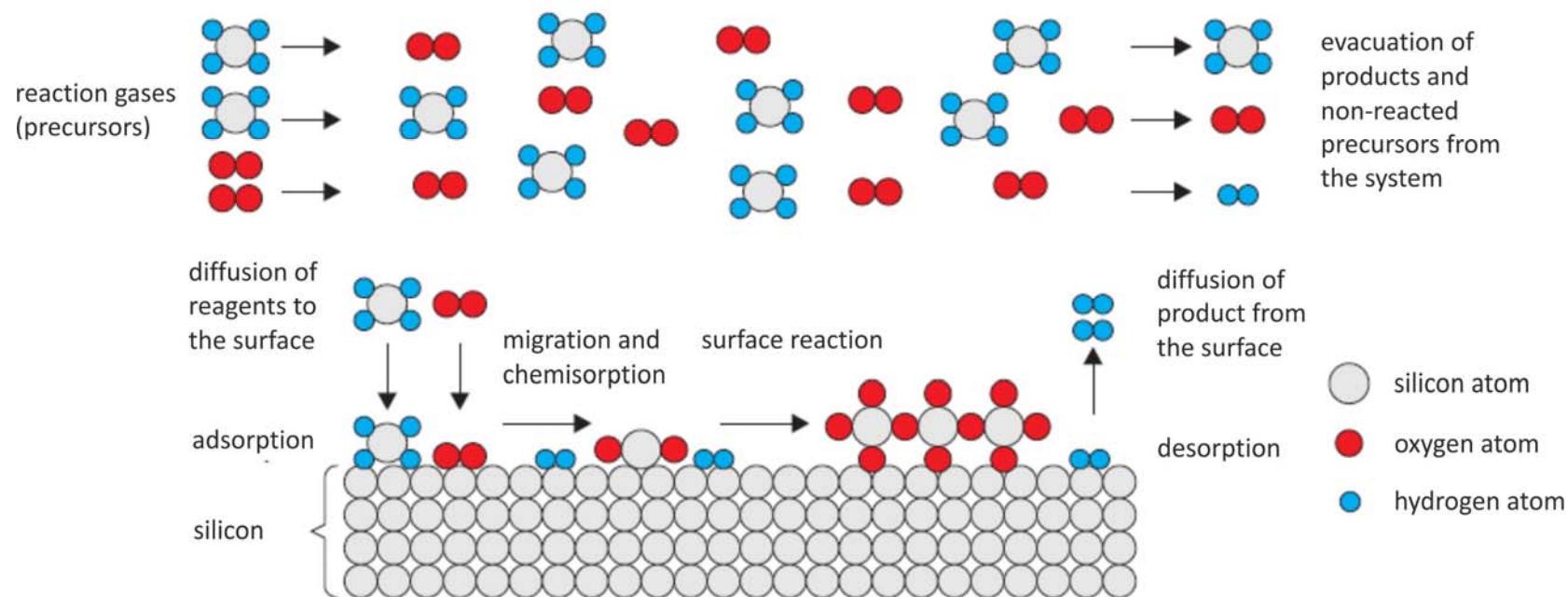
MWCVD (MicroWave...)

PACVD (Photon assisted..., or plasma assisted)

ALCVD (Atomic Layer.. ALD(ep..), ALEpitaxy)

# CVD – CHEMICAL VAPOUR DEPOSITION

- Chemical reaction of one or more gas phase reagents (precursors) on a solid substrate
  - Surface catalysed reaction (not in the gas space)
  - Solid product



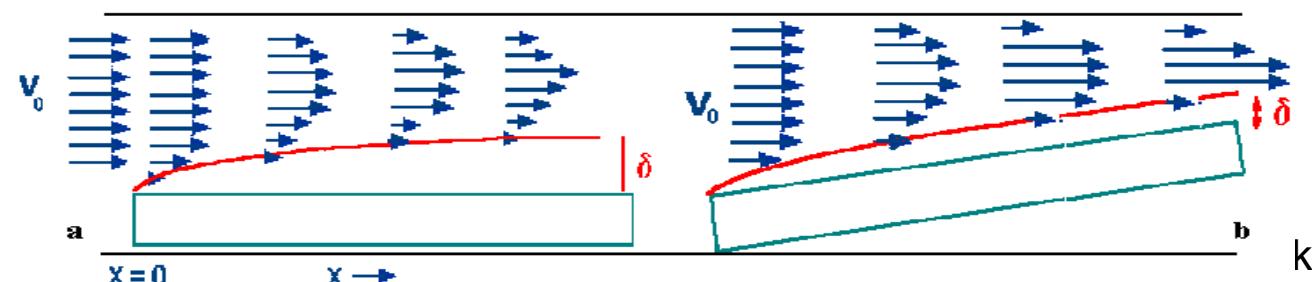
## Atmospheric CVD - APCVD

- Small free path
- Reaction rate control: transport (reagent or product)
- Thermal activation

$$\delta(x) = (\mu x / \rho v_0)^{1/2}$$

$\mu$  kinematic viscosity

$\rho$  density

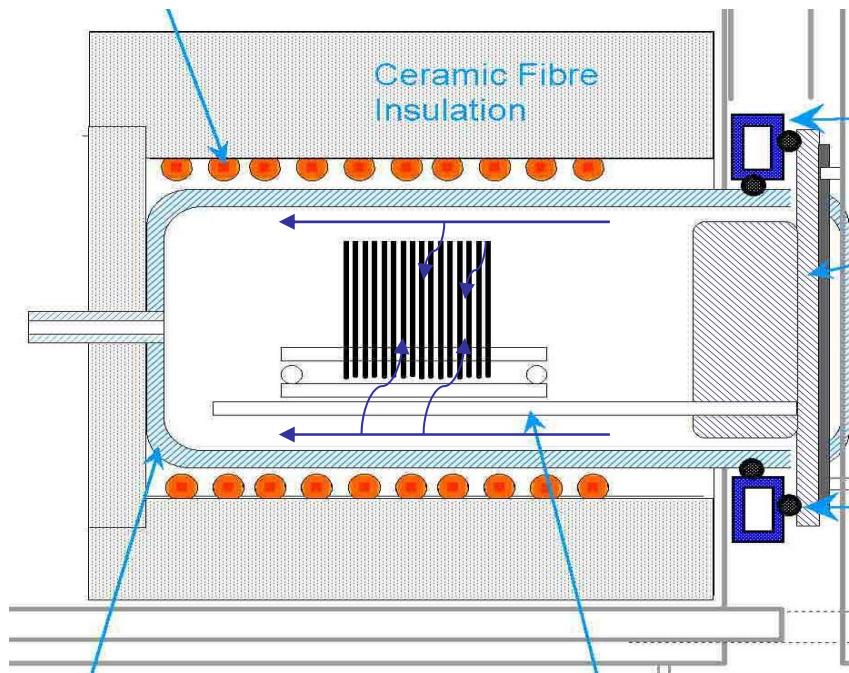


Thickness homogeneity  $\pm 10\%$ , single wafer reactors

- $\text{SiO}_2$ : silane + oxygen / 450°C

# LOW PRESSURE CVD - LPCVD

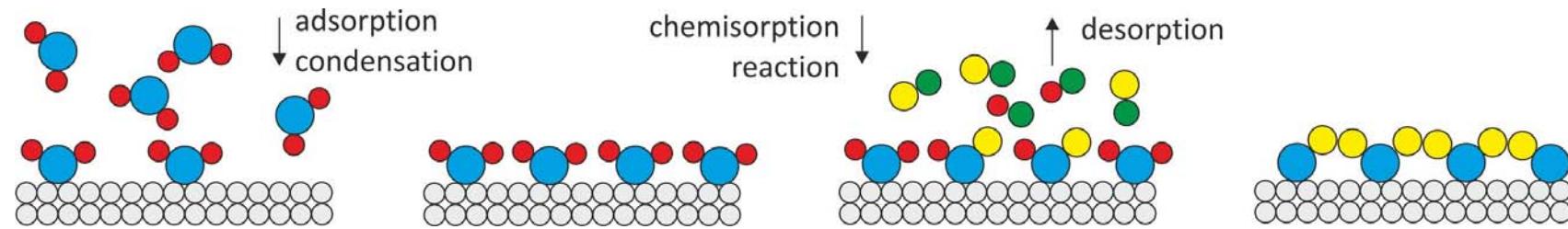
- Long free path
  - Reaction rate control: chemical reaction
  - Thermal / plasma activation



Thickness homogeneity  $\pm 2\text{-}6\%$ , batch and single wafer reactors

## ALD – ATOMIC LAYER DEPOSITION

- Reaction rate control: chemisorption
- Thermal / plasma activation



- atomic / molecular precision
- excellent homogeneity
- excellent step coverage
- batch and single wafer reactors

Typical materials:  $\text{Al}_2\text{O}_3$ ,  $\text{ZnO}$ ,  $\text{HfO}_2$ , ...



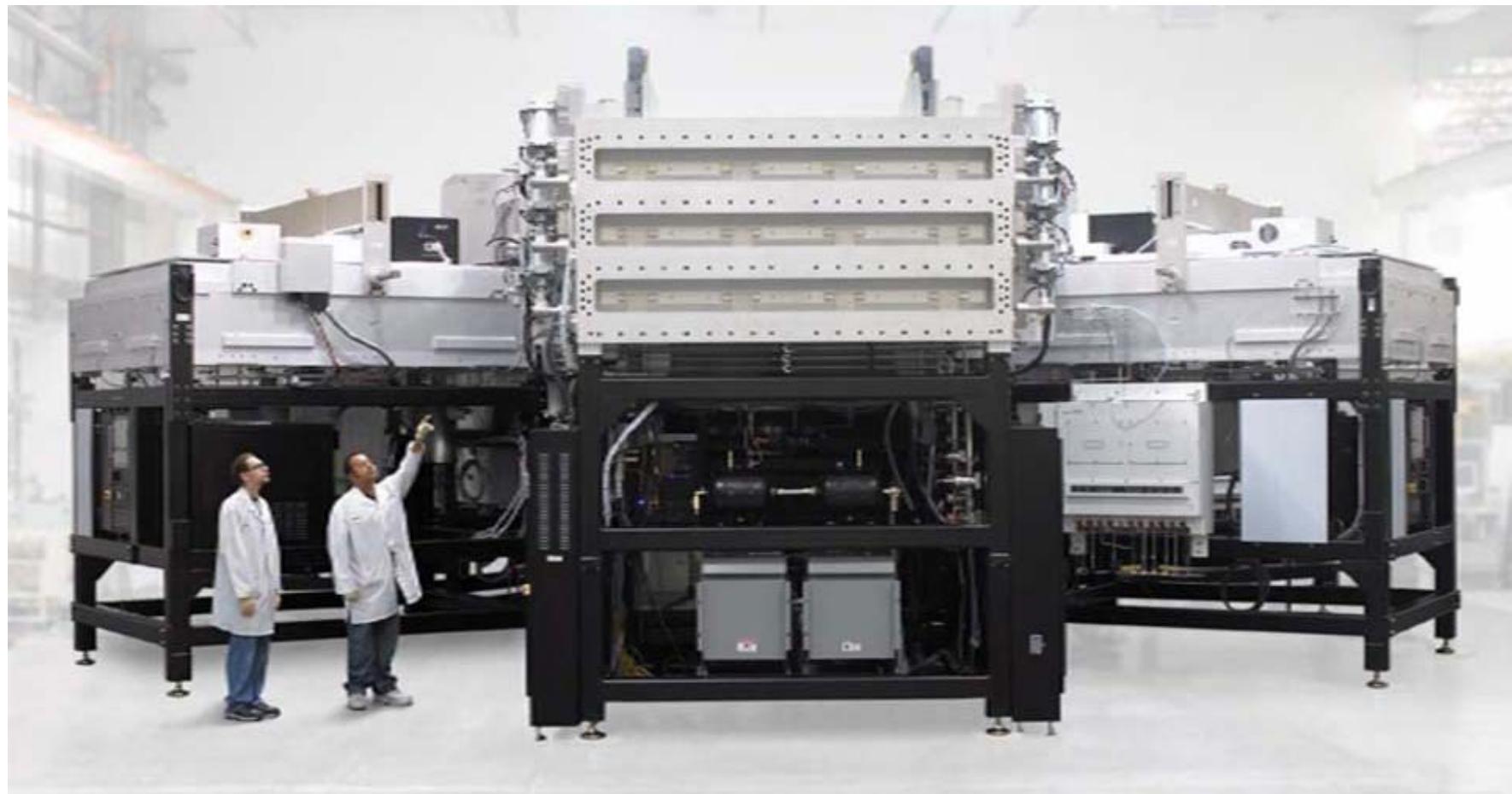


# CHEMICAL VAPOUR DEPOSITIONS

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## INDUSTRIAL SOLUTIONS



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# ETCHING

## CHEMICAL ETCHING

**Etching:** removal of the solid material of the substrate by chemical reaction

**Reagent:** liquid or gas (or vapour, or plasma)

**Wet etching:**

- **Chemical** reaction on the liquid / solid interface – causing dissolution of solid material

**Dry etching:**

- Gas or vapour phase reagents at high temperature
- Gas phase reagent at **low temperature and pressure, active particles with extreme high reactivity, generated by RF induced plasma** discharge (free radicals or excited neutral particles) – isotropic etching
- **Physical etching** – non or moderate selective **sputtering** of the substrate atoms and molecules – directional / anisotropic etching

# APPLICATIONS in IC TECHNOLOGY

## Semiconductor wafer processing

- Elimination of mechanical defects by chemical polishing
  - High quality surface development by chemical-mechanical polishing

## **CMOS technology / micromachining**

- Photoresist development
  - Selective or total removal of oxides or nitrides
  - Patterning of metal layers
  - Selective or total removal of organic layers
  - Contour etching: engineered undercut profiles
  - Anisotropic etching of si in MEMS structures
  - Etching of polycrystalline Si in MOS structures (poly-gate)

## **Analitical applications:**

e.g. exploring fouls (pinholes, crystalline fouls



**Packaging semiconductor devices:** e.g. refreshing metal surfaces



## WET CHEMICAL ETCHING

### Requirements against the etching processes:

- uniform etch rate on the whole substrate surface
- high selectivity for the masking layer  
(for photoresist or other layer)
- high selectivity for substrate material  
( $v_{layer} / v_{substrate} > 10..100$ )
- adequate etch rate corresponding to the thickness of the layer to be etched ( $\approx 0,1\text{-}1 \mu\text{m}/\text{min}$ )
- possibly controlled by chemical reaction  
(not by transport)

## WET ETCHING TECHNIQUES

### Immersion etching

- High wafer number / economical
- Rate control: temperature / stirring (bubbleing / stirring / ultrasonic tub)

### Spray etching

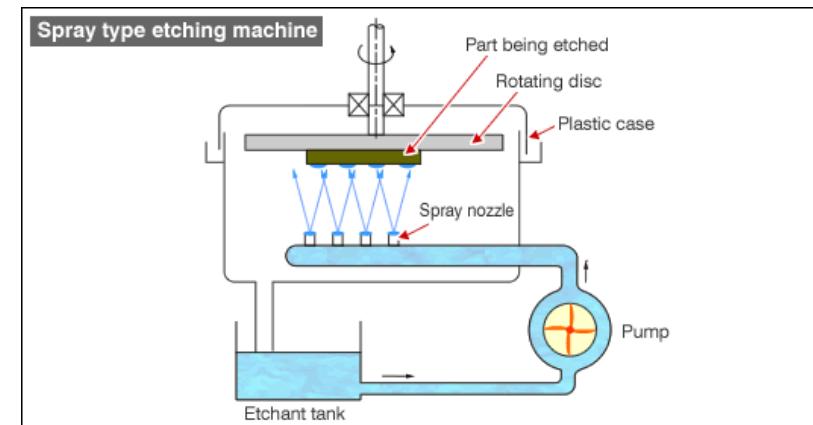
- Effective etch rate control (parameters: vaporisation drop size / pressure)
- Enhanced etch rate due to the continuously fresh etchant
- Single wafer

### Chemo-mechanical etching

- Wafer polishing (Si or polymers)

### Electrochemical etching

- Selectivity and etch rate control (parameters: potential or current)



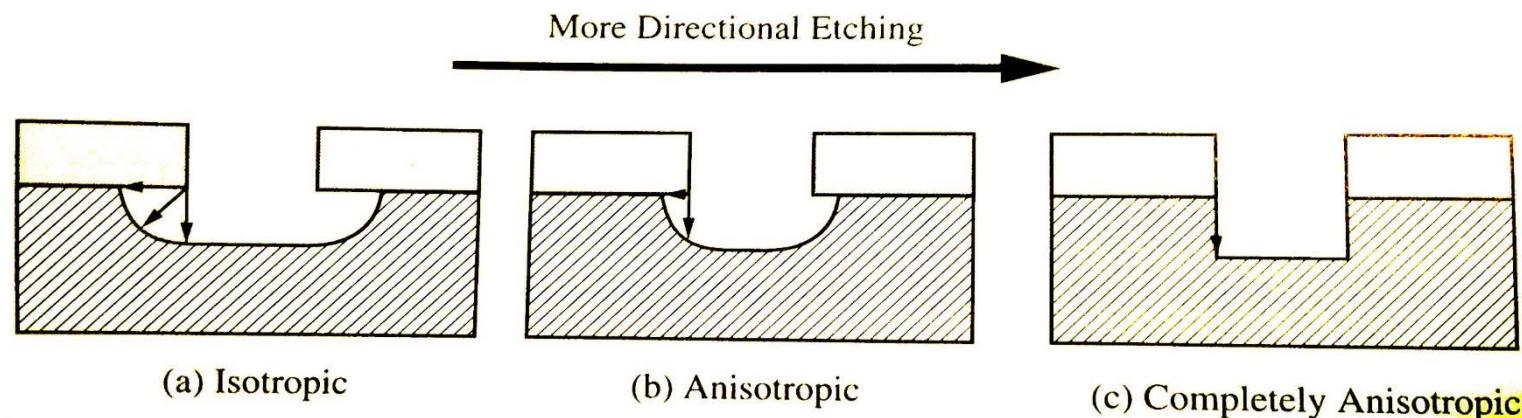
## DIRECTION DEPENDENCY OF WET ETCHING

### Isotropic etching: direction independent etch rate

- Etching of amorphous and polycrystalline materials is typically isotropic
- Typically diffusion limited processes

### Anisotropic etching: direction dependent etch rate

- Etching of crystalline materials could be isotropic and anisotropic according to the composition of the etching solution and the reaction kinetics
- Typically reaction limited processes



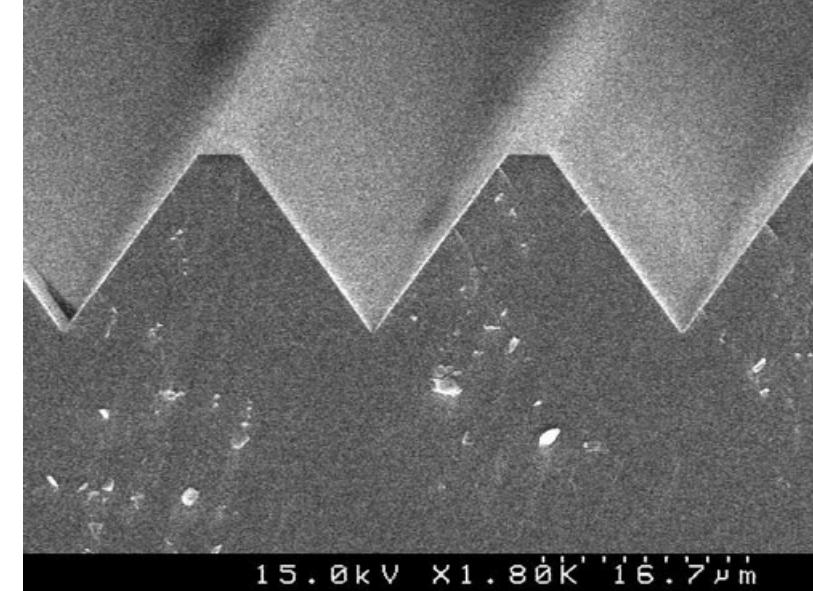
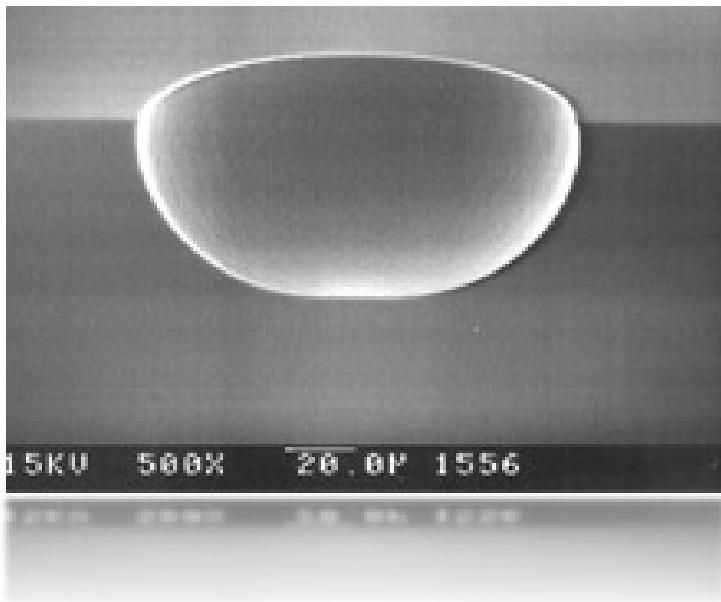


## WET ETCHING

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### ETCHING OF SILICON



#### Isotropic:

uniform etch rate in each crystallic directions  
(e.g. poly-Si etchant - HF-HNO<sub>3</sub>-CH<sub>3</sub>COOOH )

#### Anisotropic:

etch rates are altering according to the  
different crystallic directions  
(e.g. alkaline etchants – KOH)

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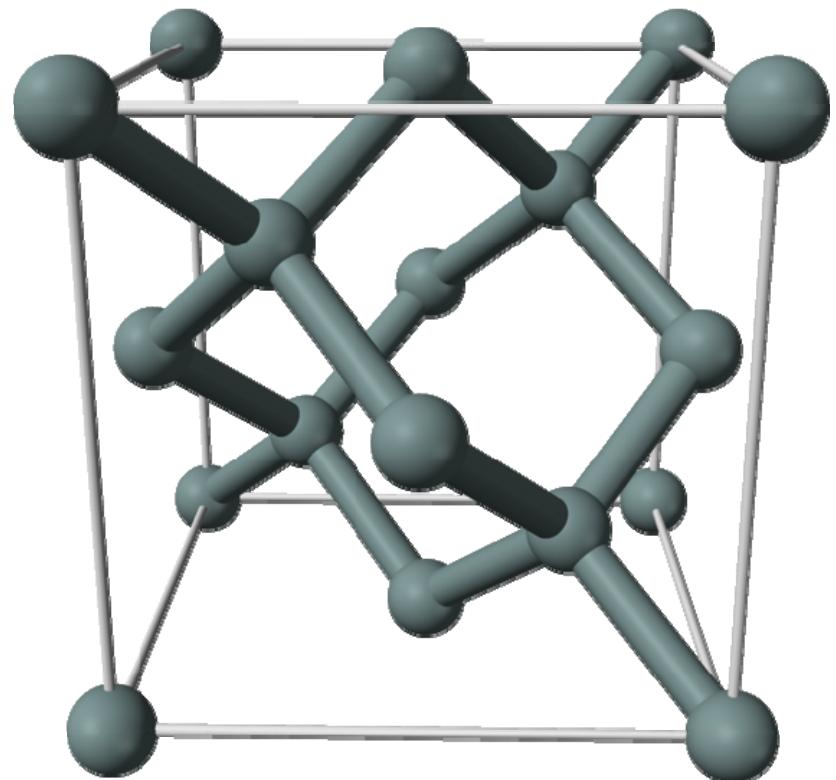
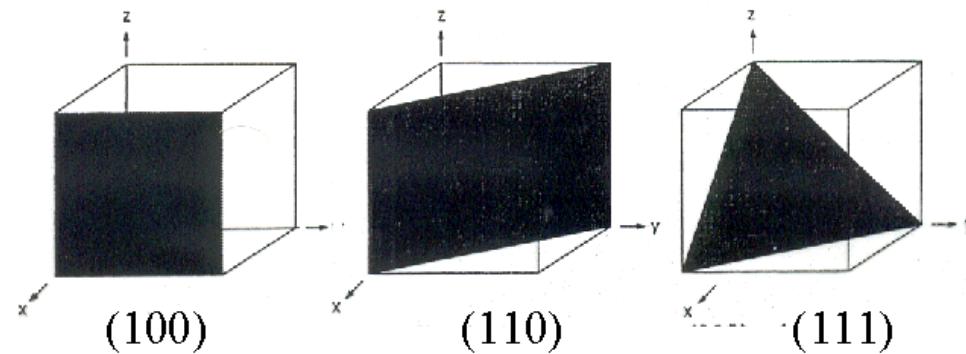
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# DIRECTION DEPENDENT ETCHING OF SI

## Crystalline structure of silicon: face centered cubic

## Typical crystalline planes:



## Si-Si bonding energies:

$$E_{\sigma(\text{SiSi})(111)} >> E_{\sigma(\text{SiSi})(100)} > E_{\sigma(\text{SiSi})(110)}$$

## Etching rates:

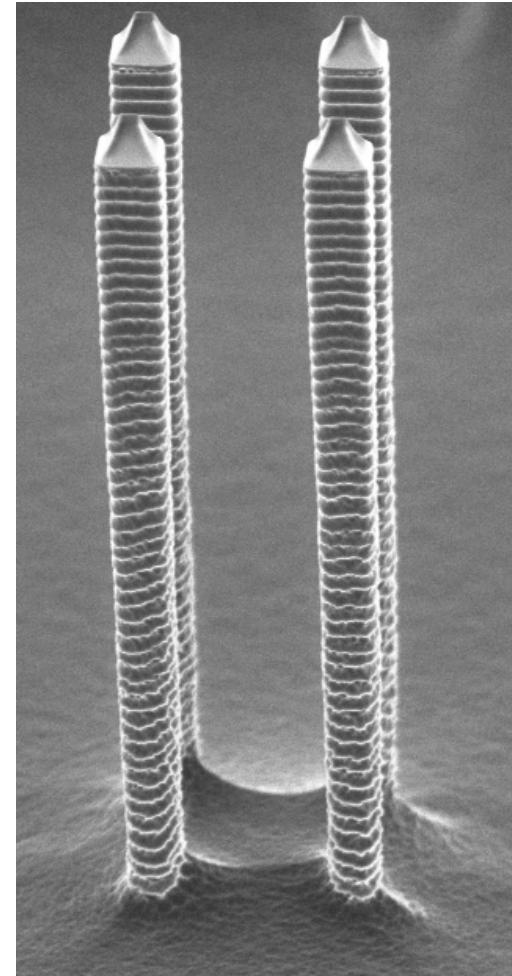
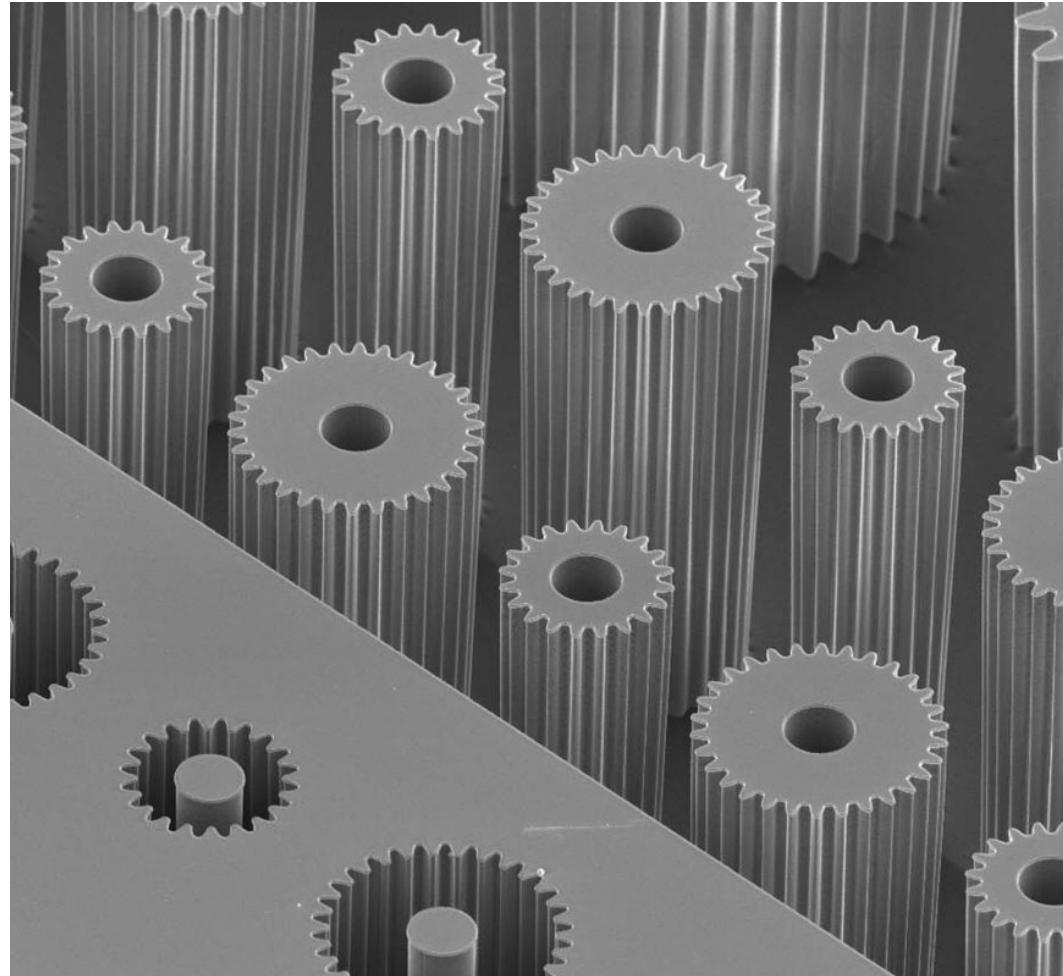
$$V_{<111>} << V_{<100>} < V_{<331>}$$



## DRY ETCHING

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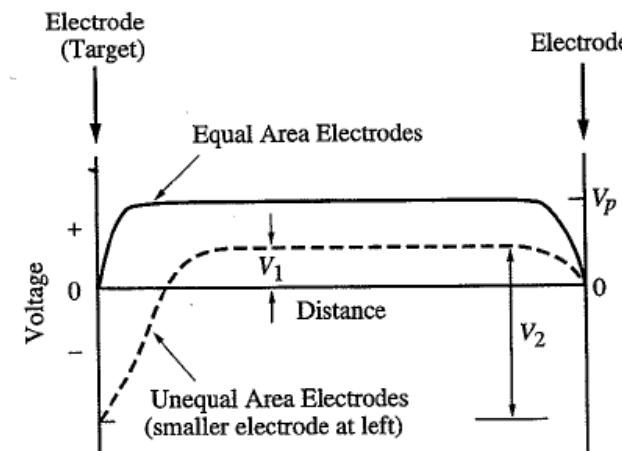
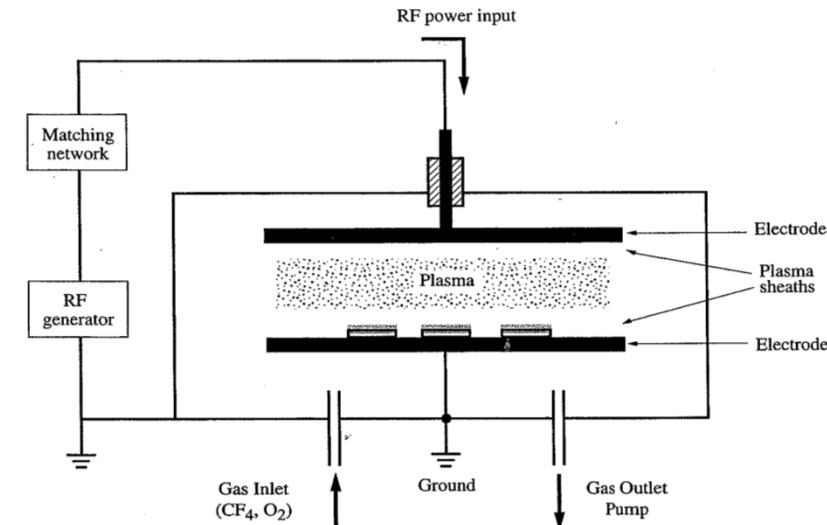
# PLASMA ETCHING

# Plasma Glow

- Low gas pressure (1 mtorr-1 torr)
  - High electric field on the electrodes, 13.56 MHz RF
  - Ionisation of the gas atoms:  
 $e^- + \text{ions}$

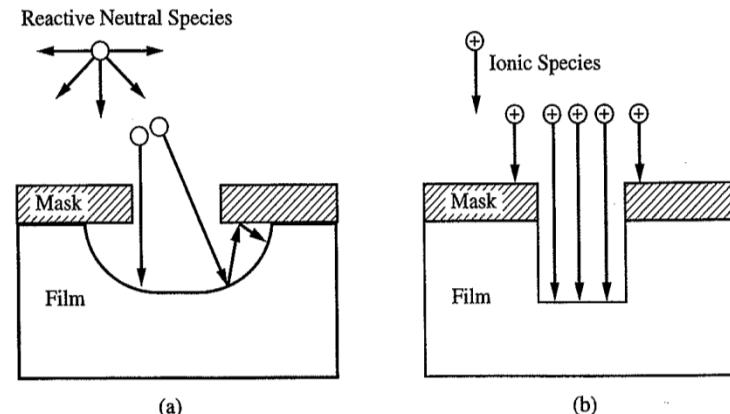
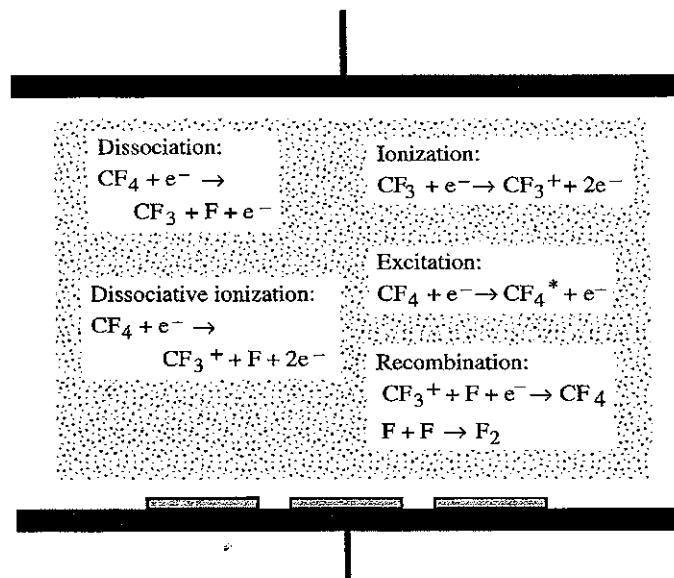


**plasma glow** – conducting gas  
(ions, free radicals, electrons, natural particles),  
Particles are excited by the quick electrons  
and emit photons after relaxation.



# DRY ETCHING PROCESSES

- Effective **chemical etching** by reactive radicals (atomic F)
  - Directional / anisotropic **physical etching** by charged particles

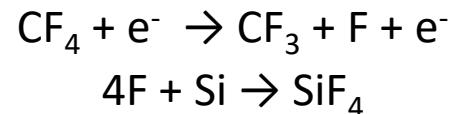


**Figure 10-11** Fluxes of species in plasma etching: (a) fluxes of reactive neutral chemical species (such as free radicals), with a wide arrival angle distribution and low sticking coefficient; (b) fluxes of ionic species, with a narrow, vertical arrival angle distribution and high sticking coefficient (assumed equal to 1).

**Figure 10-9** Typical reactions and species present in a plasma used for plasma etching.

# DRY CHEMICAL ETCHING

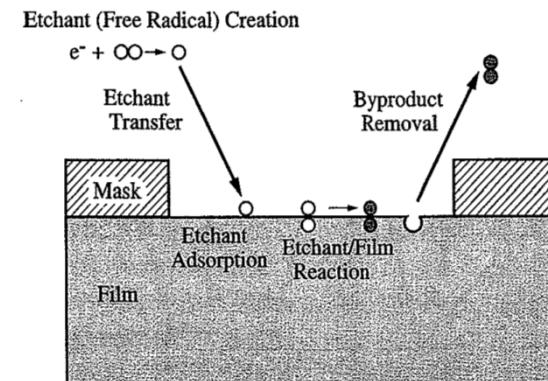
Free radicals (neutral, having non-bonding electron pair) – extremely reactive



**Volatile products** – must get away from the surface  
for continuous etching

**Additive gases:** possibly support the generation of reactive free radicals, enhancing etch rates!

e.g.  $O_2$  gas reacts with dissociated  $CF_3$ ,  $CF_2$  molecules, preventing the recombination to  $CF_4$ , enhancing the concentration of free F radicals  
BUT:  $O_2$  dilutes the etchant gas!



**Figure 10-10** Processes involved in chemical etching during plasma etch process.

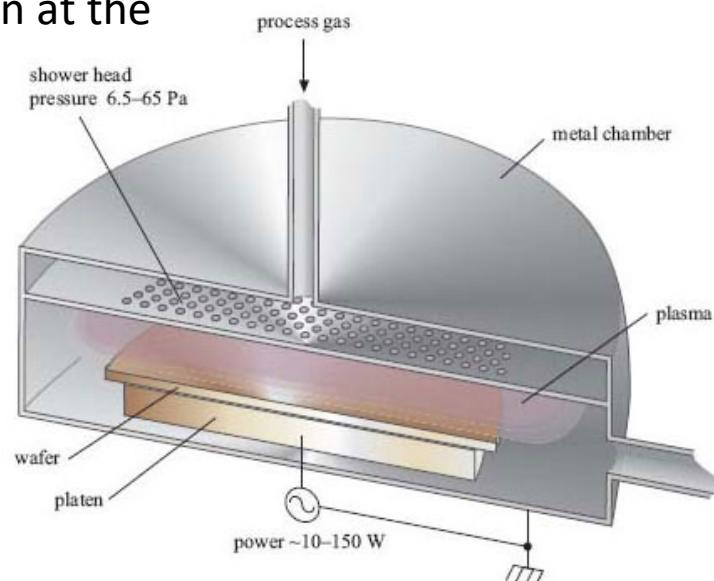
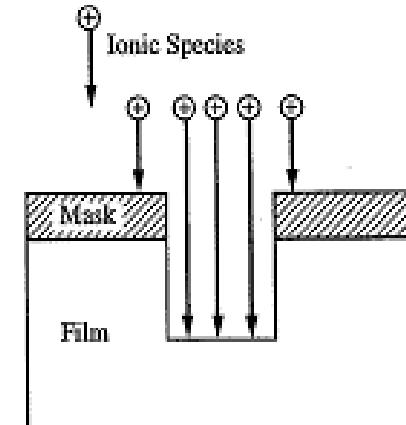
## Isotropic etching:

1. Isotropic angular distribution of the incident velocity vector (particles)
  2. Low surface adhesion / sticking coefficient  
(long path till reaction)

## HIGH SELECTIVITY

## **DRY PHYSICAL ETCHING**

- **Positive ions are accelerating towards the electrodes** due to  $V_p$  (one is the substrate holder)
  - Anisotropic etching:
    - Direction dependent etching rate of the incident ions due to the directional electric field
    - High adhesion / sticking coefficient – reaction at the moment of incidence
  - **LOW SELECTIVITY**



## Technologies:

- Sputtering or ion etching
  - Focused Ion Beam etching (FIB)
  - Magnetically localised ion etching

## ION-ASSISTED ETCHING

## **Chemical-physical dry etching**

**(combination of the two processes)**

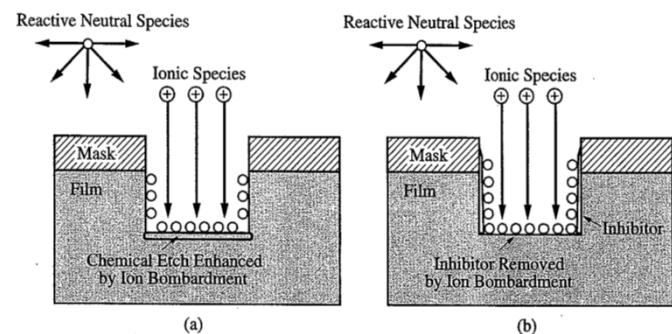
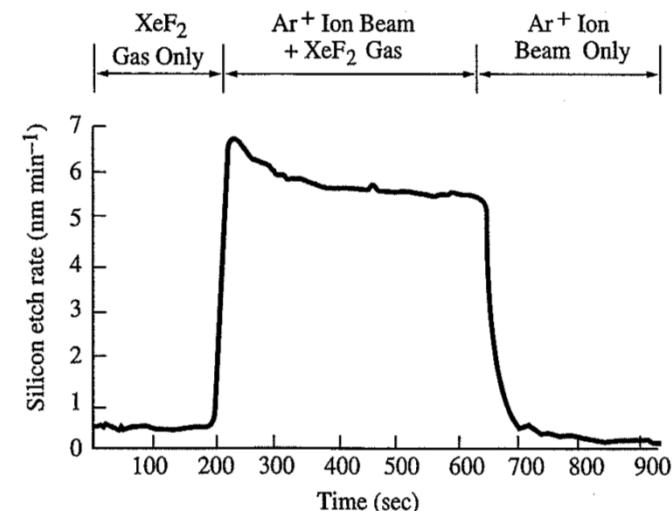
Ions + natural free radicals etch dependently:

- Can increase **selectivity and orientation dependent reaction rate**
  - The etch rate is not the sum (higher)
  - The etch profile is not a linear combinatin, but similar to physical etching (vertical etch rate increases)

**The ion bombardment enhances one of the component of the chemical etching (surface adsorption, etching reaction, generation / removal of the product) anisotropic way**

## Technics:

- Reactive ion etching, sputtering
  - Reactive ion beam etching
  - Chemical enhanced ionbeam etching



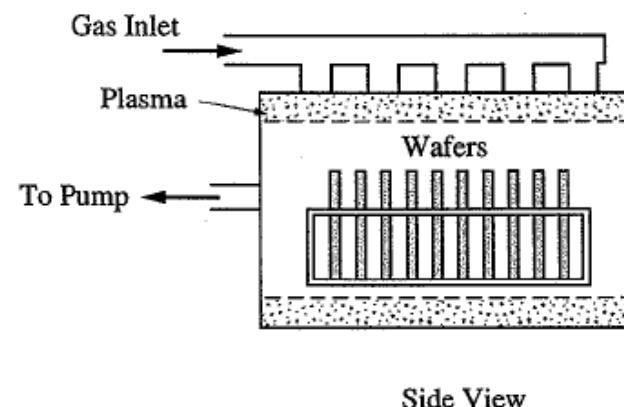
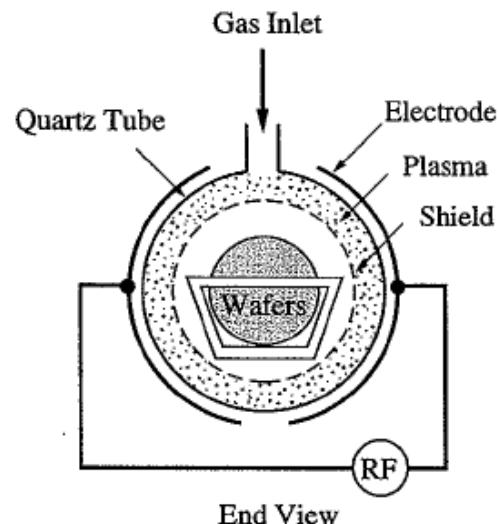
**Figure 10-13** Illustration of ion-enhanced etching. In (a) the chemical etch reaction is enhanced by ion bombardment. In (b) an inhibitor is formed which is removed by ion bombardment, allowing chemical etching to proceed. In both cases, anisotropic etching results.

# PLASMA ETCHING EQUIPMENTS I.

# Cylindrical / barrel type plasma etcher

- Wafer in holder (not on the electrode), multiwafer process
  - Isotropic chemical etching, high selectivity, low fault generation
  - Inhomogeneous etch rate on the wafer
  - $p=10-1000\text{ mtor}$

For not critical  
etching steps  
*resist removal*  
*in O<sub>2</sub> (ashing)*

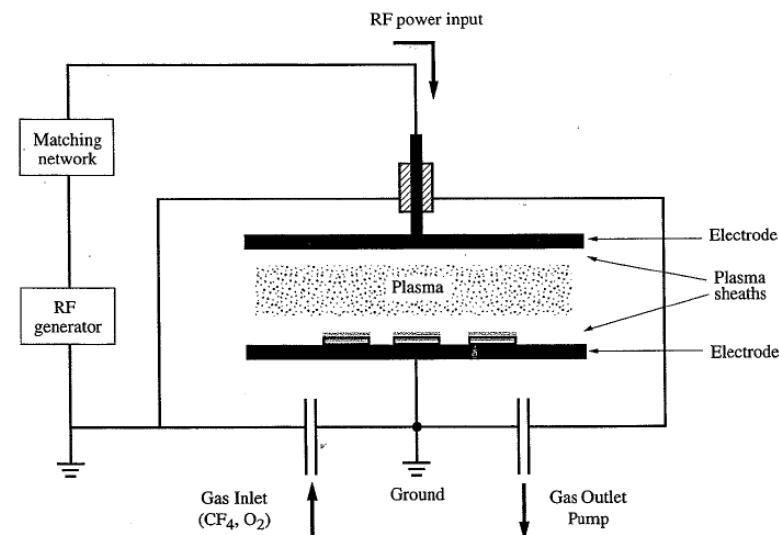


# PLASMA ETCHING EQUIPMENTS II.

## **Planar type plasma etcher - Plasma mode**

- The wafer is on the (bigger) grounded electrode facing to the opposite electrode – higher homogeneity, mainly chemical, adequate selectivity, slight anisotropy
  - Weak ion bombardment , potential difference 10-100V
  - The smaller electrode is sputtered
  - $p=10\text{-}500\text{ mtorr}$
  - Ion concentration  $\sim 10^9\text{-}10^{10}\text{ cm}^{-3}$

For not critical etching steps  
*resist removal in O<sub>2</sub> (ashing)*  
*Isotropic silicon-nitride etching*



**Figure 10-7** Schematic diagram of an RF-powered plasma etch system.

# PLASMA ETCHING EQUIPMENTS III.

## **Planar type plasma etcher – RIE (Reactive Ion Etching) mode**

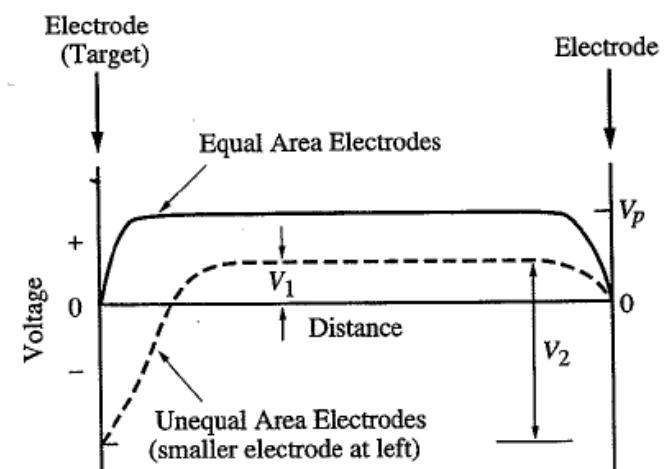
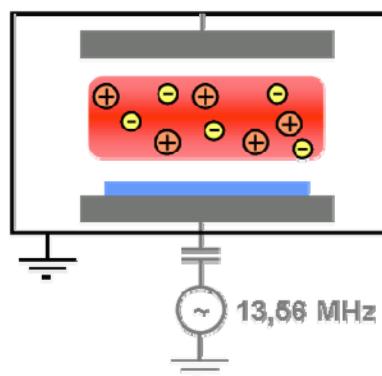
- The wafer is on the smaller electrode – single wafer process
  - The bigger electrode is grounded and connected to the chamber wall, higher potential difference in the range of 100-800V (bias) - ion enhanced / assisted anisotropic etching
  - More directional etch in case of low pressure, but lower plasma density (10-100 mtorr), ion concentration  $\sim 10^9\text{-}10^{10}\text{ cm}^{-3}$
  - Moderate etching rate 100 nm/min
  - Lattice faults, charging, trenching

## Examples:

SiO<sub>2</sub>: CHF<sub>3</sub>

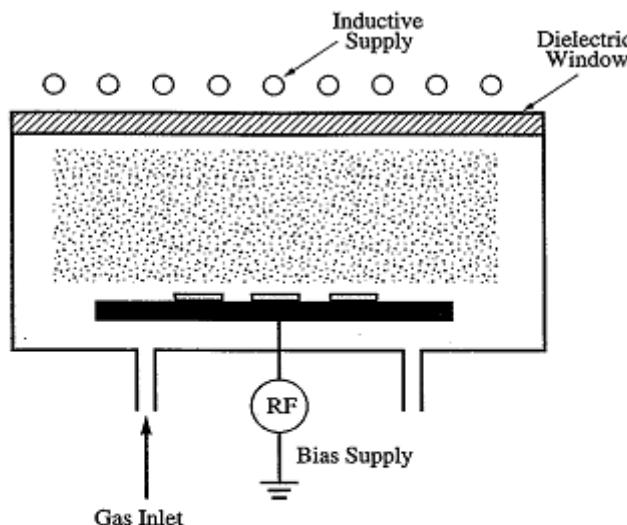
poly-Si, Si<sub>3</sub>N<sub>4</sub>: SF<sub>6</sub>+O<sub>2</sub>, NF<sub>3</sub>

Al: Cl<sub>2</sub>, BCl<sub>3</sub>



# **PLASMA ETCHING EQUIPMENTS IV.**

## **HDPE - High Density Plasma Etching**



**Figure 10-16** Schematic diagram of High-Density Plasma (HDP) etch system. This configuration is powered by an Inductively Coupled Plasma (ICP) source which produces and controls the high-density plasma. The RF wafer bias independently controls the ion energy.

- **Independent plasma density and ion energy**
  - ECR (electron-cyclotron-resonance) or ICP (inductively coupled plasma) source generates  $10^{11}$ - $10^{12}$  ion/cm<sup>3</sup> plasma density, without high sheath bias – lower pressure can be applied 1-10 mTorr – highly directional (less collision in the sheath)
  - RF source develops the potential difference, defines the bombarding ion energy, (can be decreased besides high ion density – decreased substrate deterioration)
  - high etch rate: some μm/min

Similar effect as in case of ion enhanced etching!

# DRIE INTRO

# DRIE – Deep Reactive Ion Etching

Etching depth / trench width > 10:1 (MEMS, DRAM capacitors)

## Doubled power sources:

- ICP to achieve extremely high density reactive radicals + ions
  - CCP DC self-bias for definition ion energies

Si DRIE

Gas composition: halogen based accelerated plasma etching

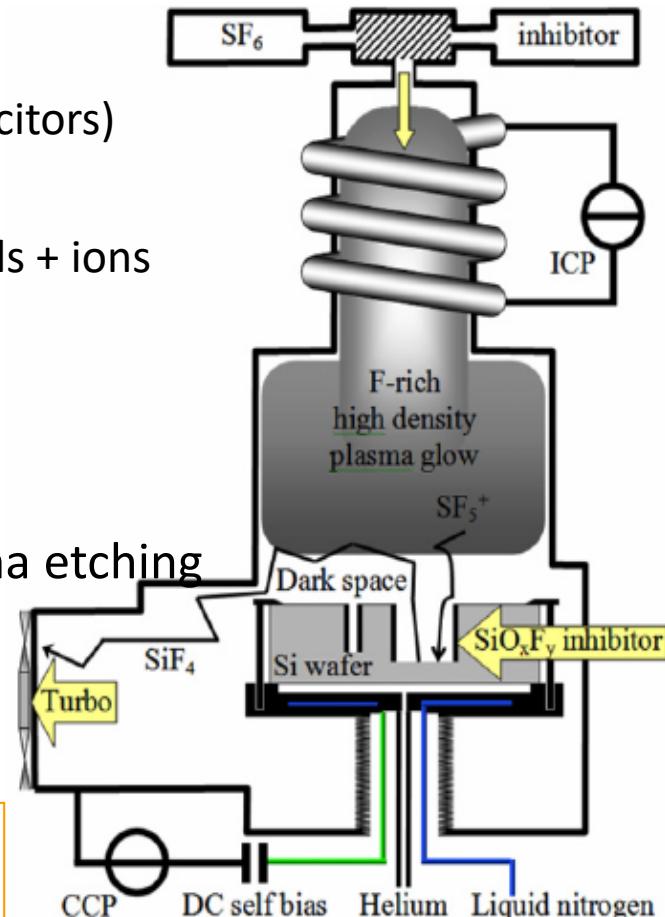
- F-based, (e.g. SF<sub>6</sub>) quick isotropic etching
  - Cl-, Br-based (e.g. Cl<sub>2</sub>, HBr) anisotropic with ion assisted etching, but slower and poisoning

## Mixed mode DRIE / Cryo

SF<sub>6</sub> + O<sub>2</sub> @ cryo °C

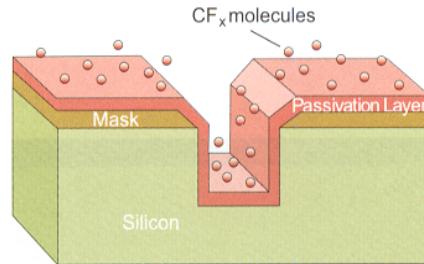
Pulsed mode DRIE / Bosch

SF<sub>6</sub> + C<sub>4</sub>F<sub>8</sub> @ RT

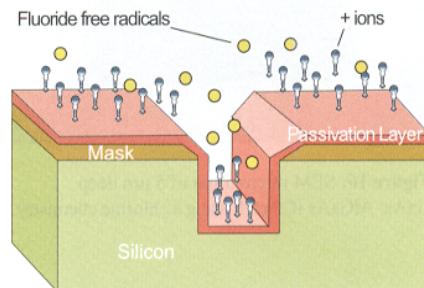
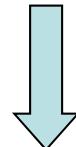


**Figure 1.** A dual source DRIE system.

## DRIE – BOSCH PROCESS



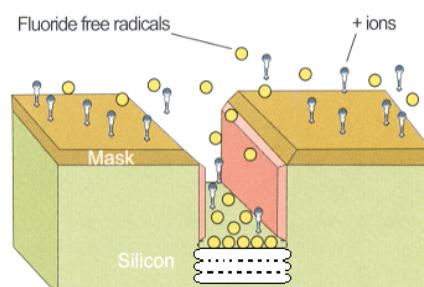
- Passivation



- Etching



ion bombardment + polymer etching  
(excluding the vertical walls)



- $\text{SF}_6$  isotropic or  
slightly anisotropic Si etching





# DRY ETCHING

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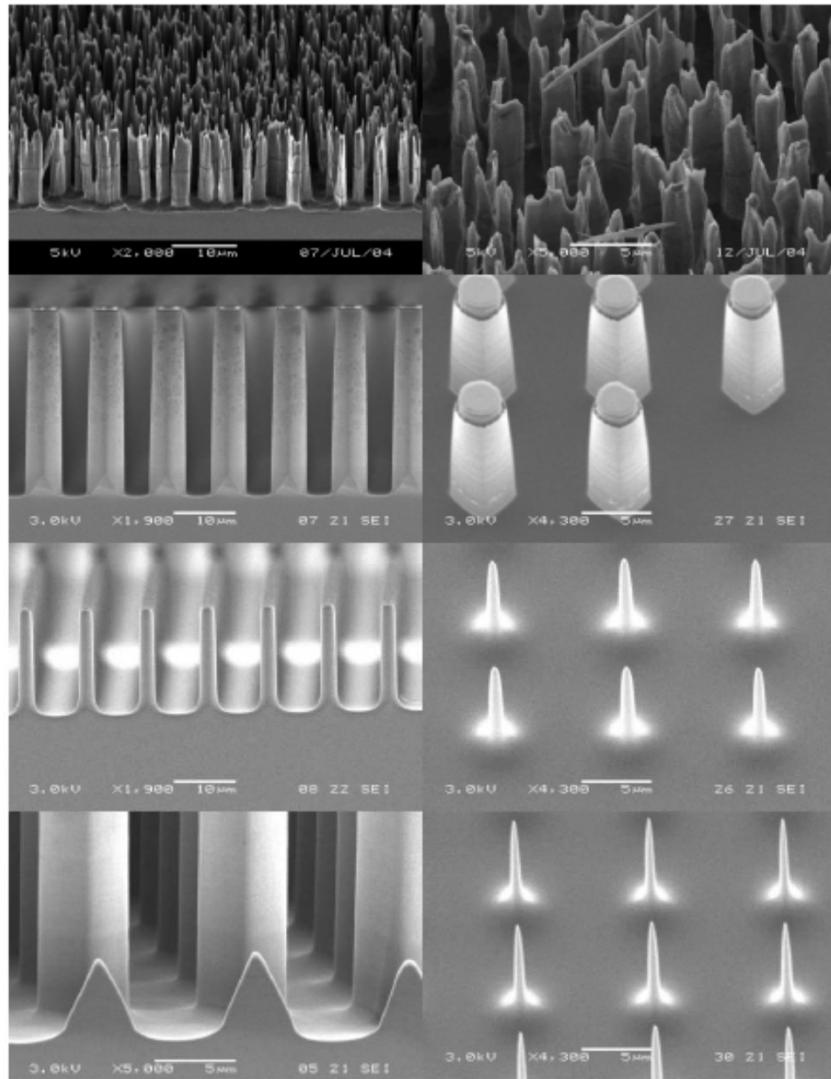


Figure 19. (Top) Black silicon and (rest) optimized result for cryogenic temperature mixed-mode DRIE (see figure 27).

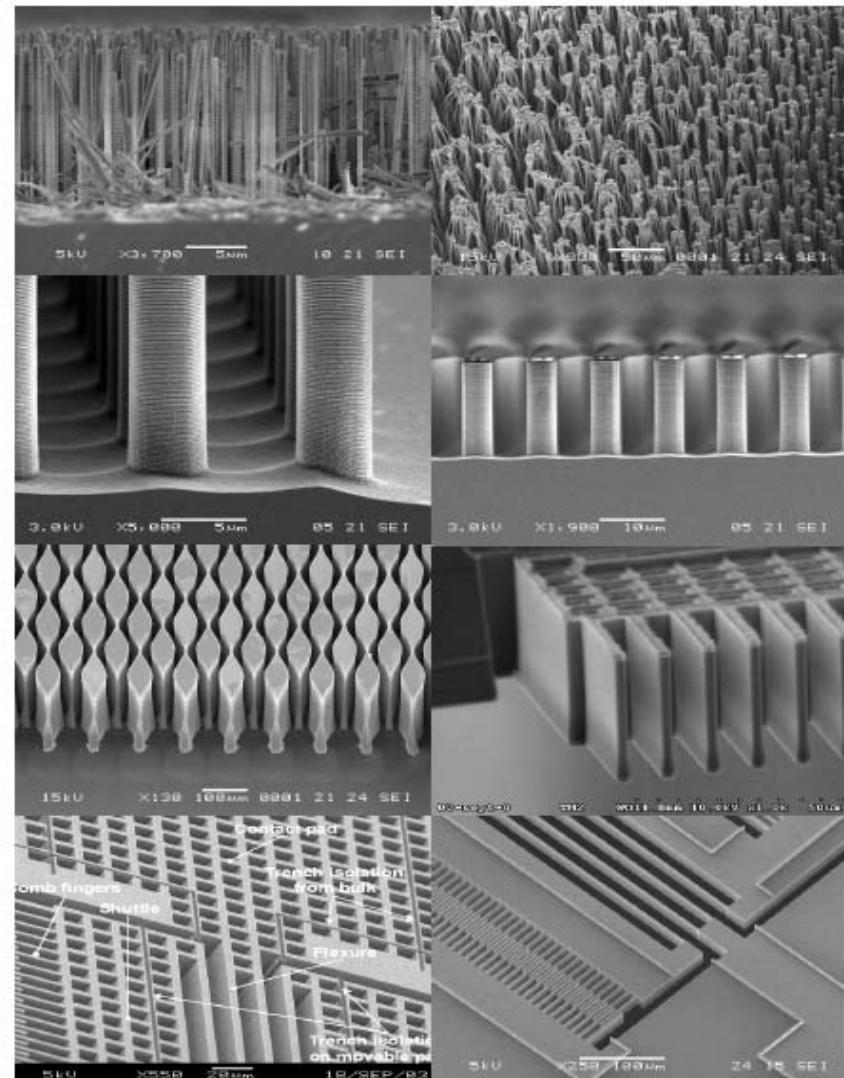


Figure 27. Typical result for room temperature pulsed-mode DRIE (see figure 19).



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# WAFER BONDING

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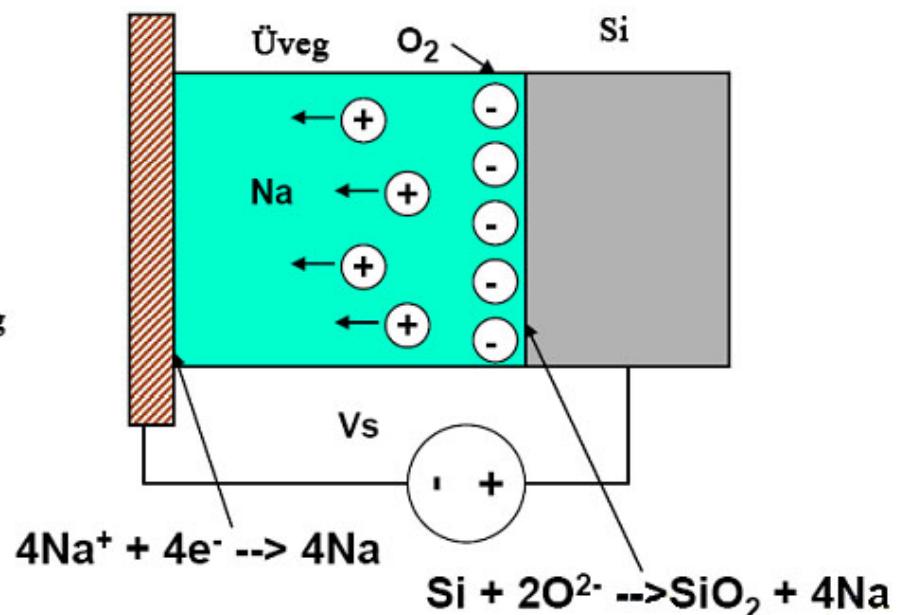
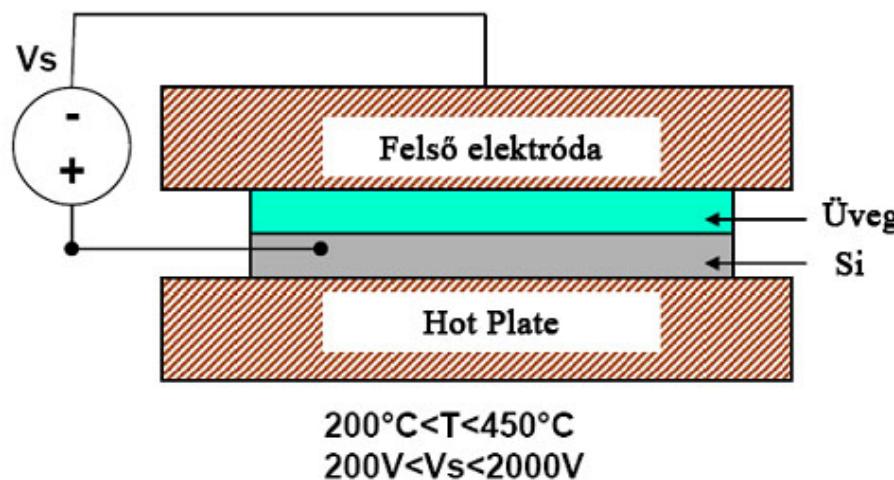
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## ANODIC BONDING

- Si + Special glass (high alkaline-ion concentration)
- Moving Na<sup>+</sup> ions – depleted space-charge layer
- Covalent bonding of silicon and oxygen
- Low sensitivity for surface roughness





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# LIGA

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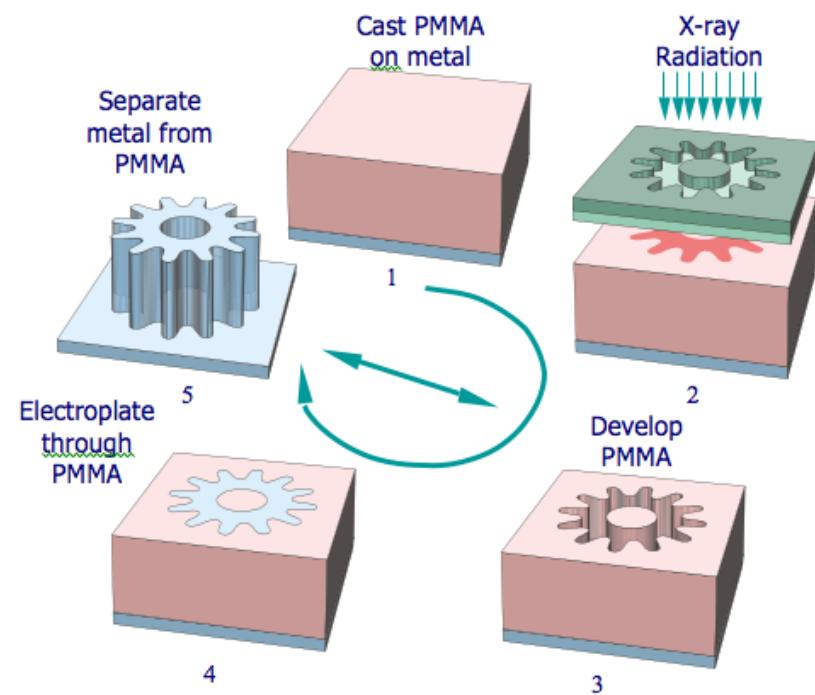
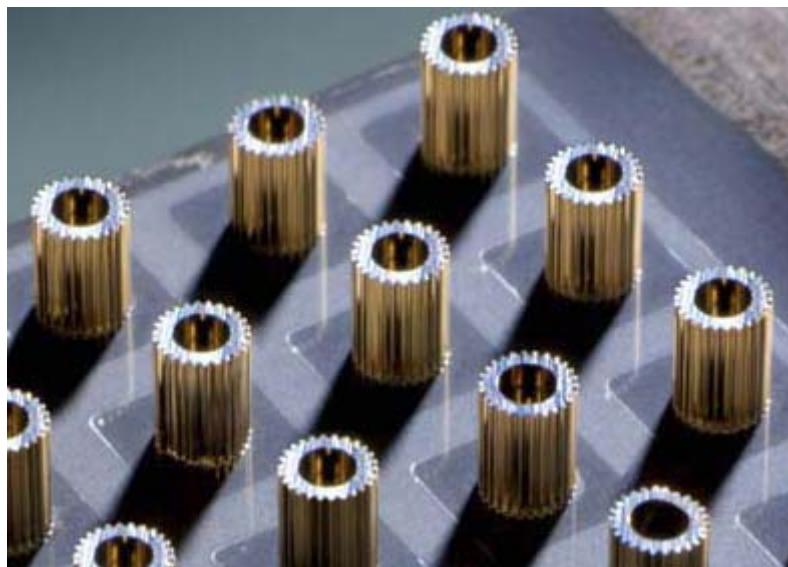
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## LIGA (KIT)

### Lithographie, Galvanoformung, Abformung (Lithography, Electroplating, Casting)

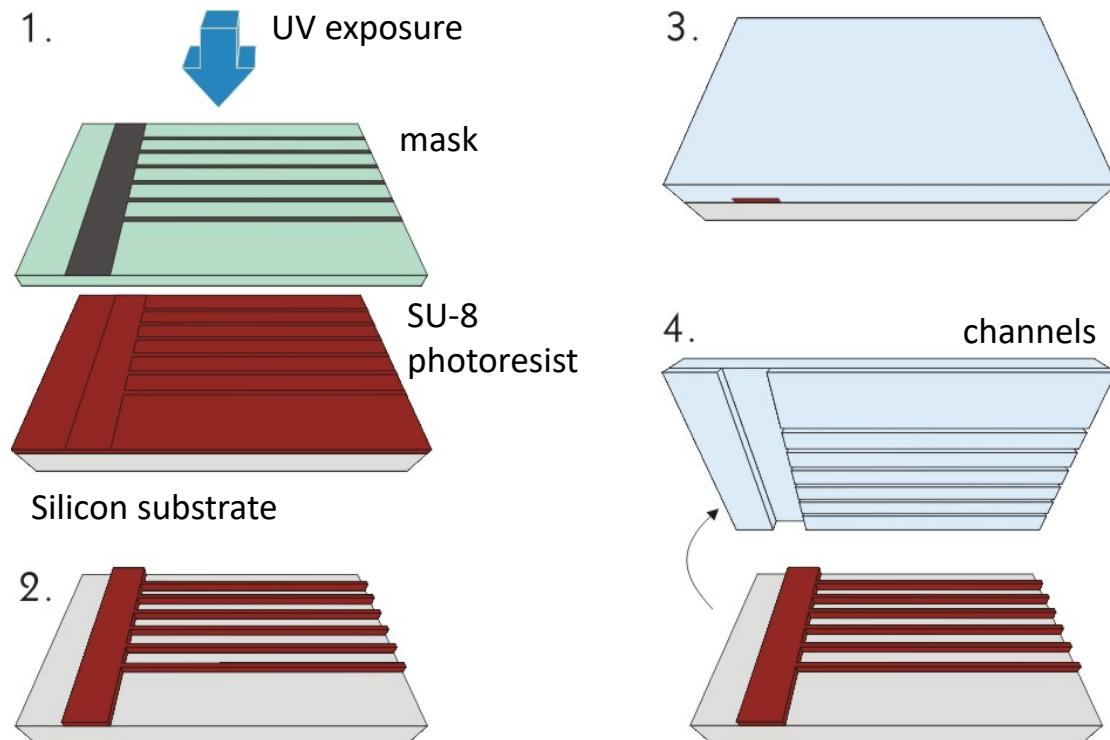
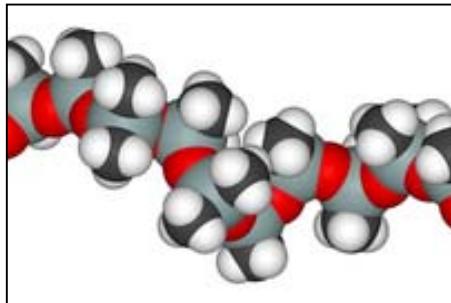
- Fabrication microstructures with high aspect ratio (100:1)
- Vertical sidewalls, 10nm surface roughness (optical structures)
- Height: from 10 $\mu\text{m}$  to some mm
- X-ray LIGA (PMMA) / UV LIGA (SU-8)





# SOFT LITHOGRAPHY

## FABRICATION PDMS POLYMER MICROFLUIDIC STRUCTURES

**Advantages:**

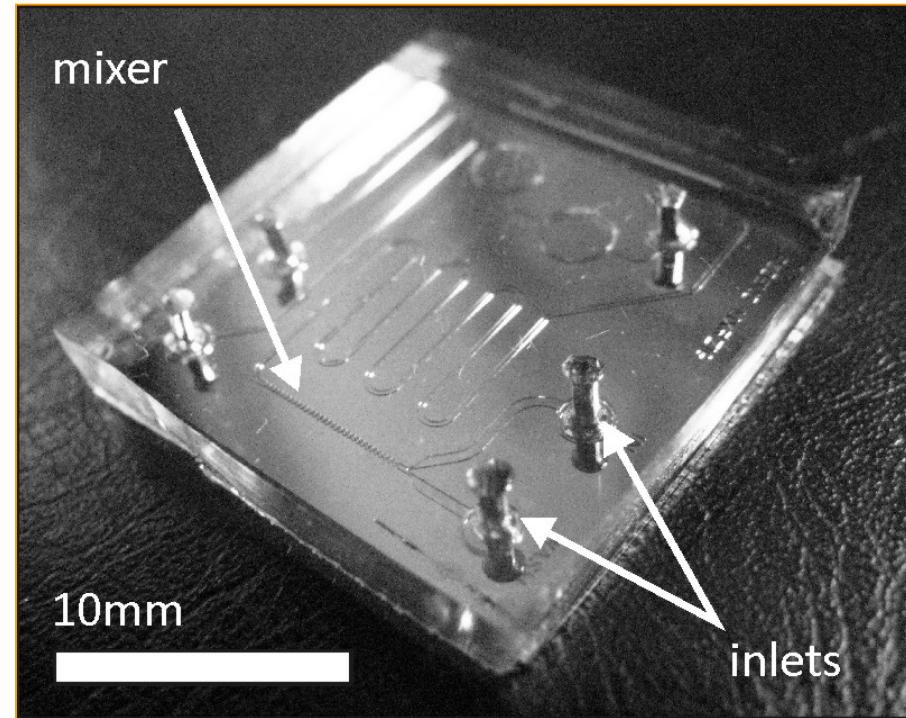
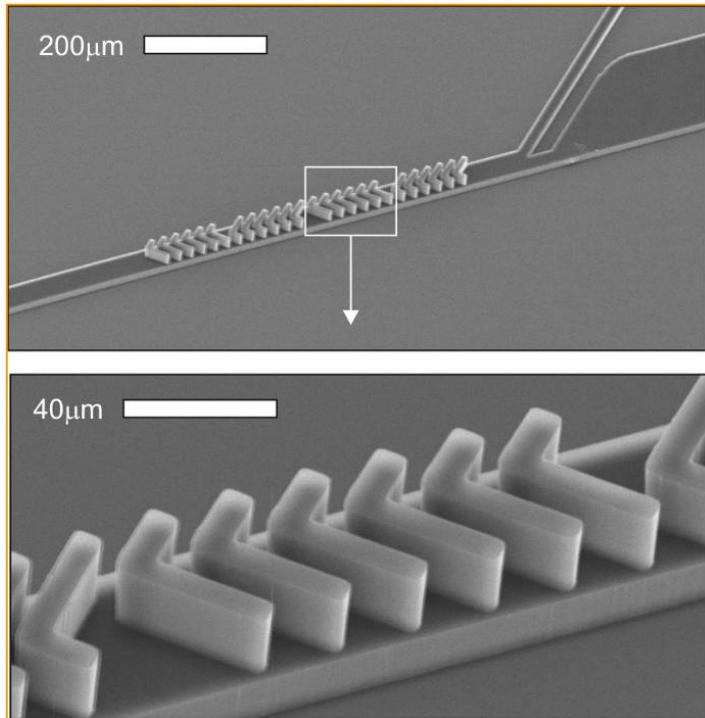
- biocompatibility, flexibility, transparency
- cheap, fast, easy to use
- covalent bonding to Si, glass and PDMS surfaces

**Disadvantages:**

- hydrophobic
- non-specific molecule (e.g. protein) adsorption

## FABRICATION PDMS POLYMER MICROFLUIDIC STRUCTURES

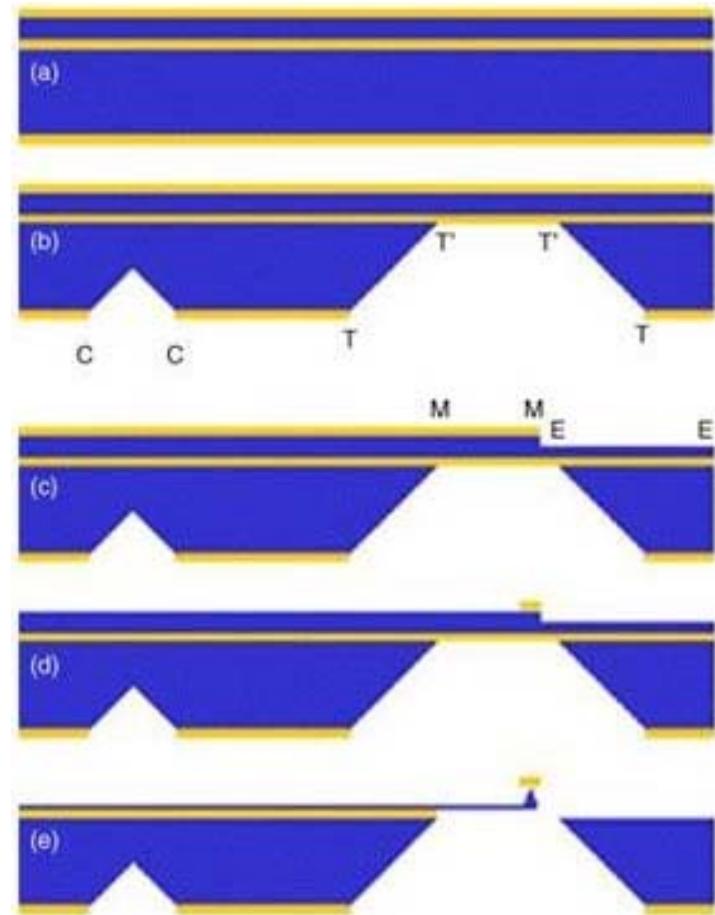
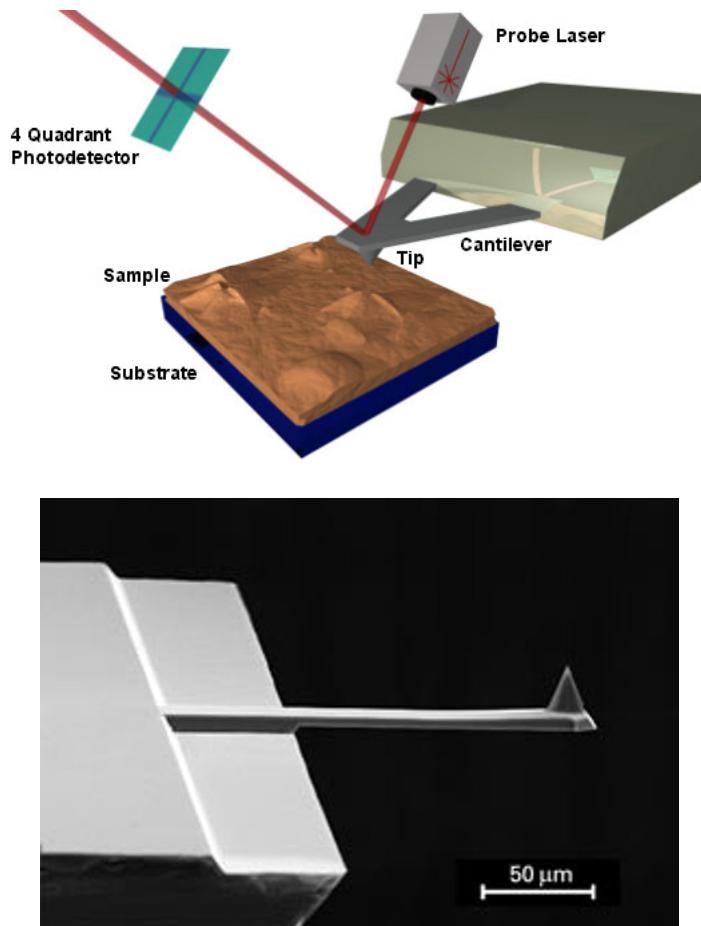
- Multi-layered 3D SU-8 technology for structuring moulding master
- FAST PROTOTYPING – PDMS moulding / casting



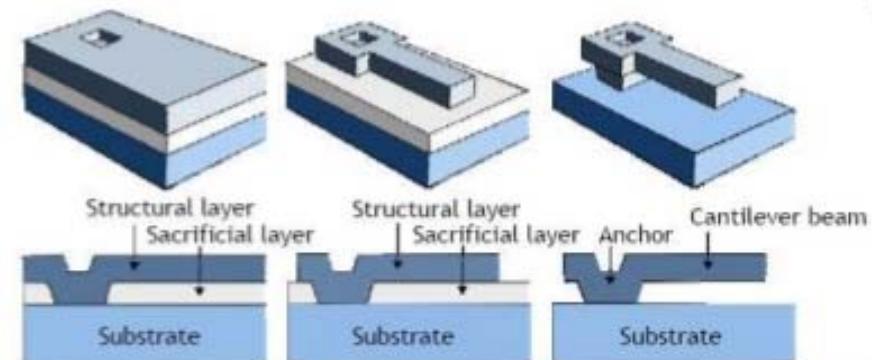
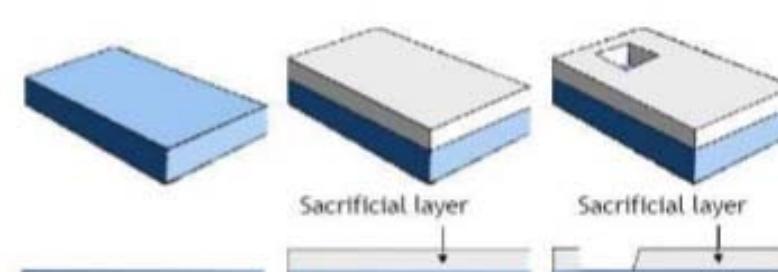
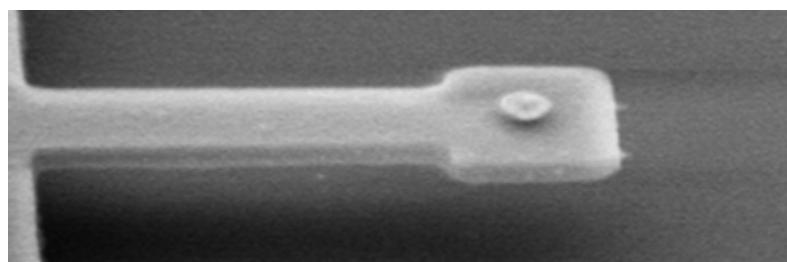
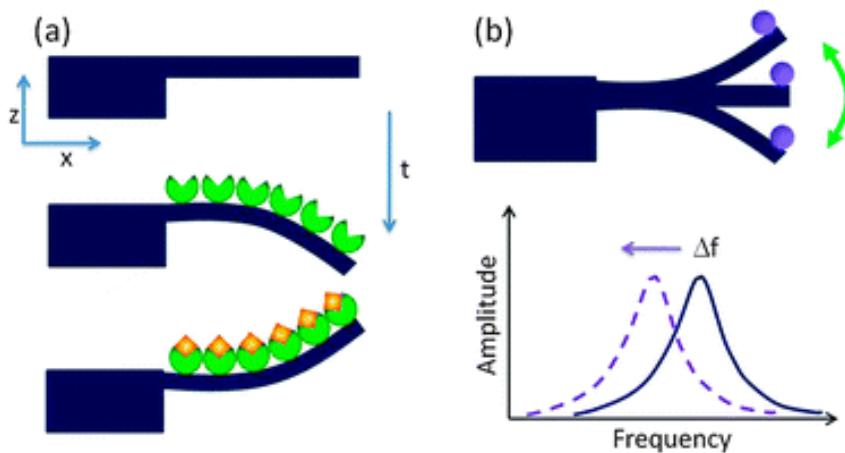
Herring-bone type chaotic mixer

# EXAMPLE DEVICES

# CANTILEVER – BULK MICROMACHINING

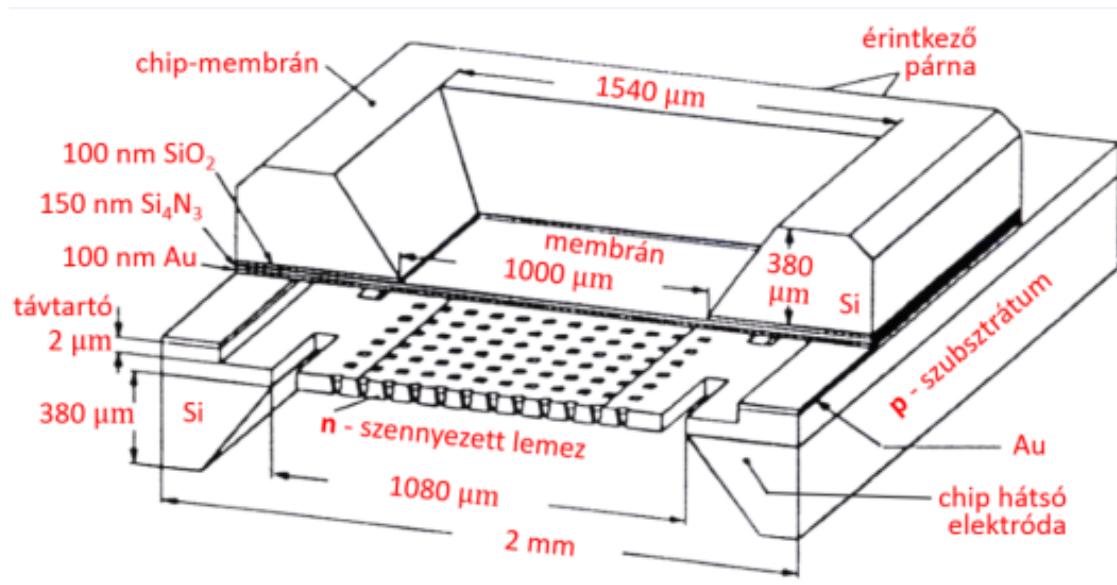


# CANTILEVER – SURFACE MICROMACHINING



## MICROPHONE

High Performance MEMS microphones (3-4 pcs / phone)

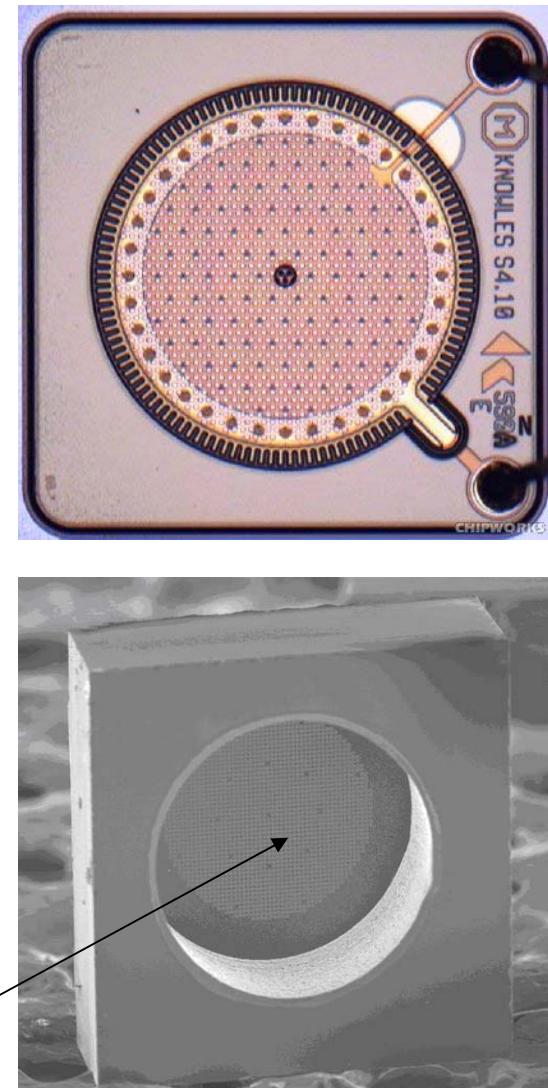


Top electrode: Au SiO<sub>2</sub> / SiN<sub>x</sub> membrane  
Bottom-electrode: n-Si

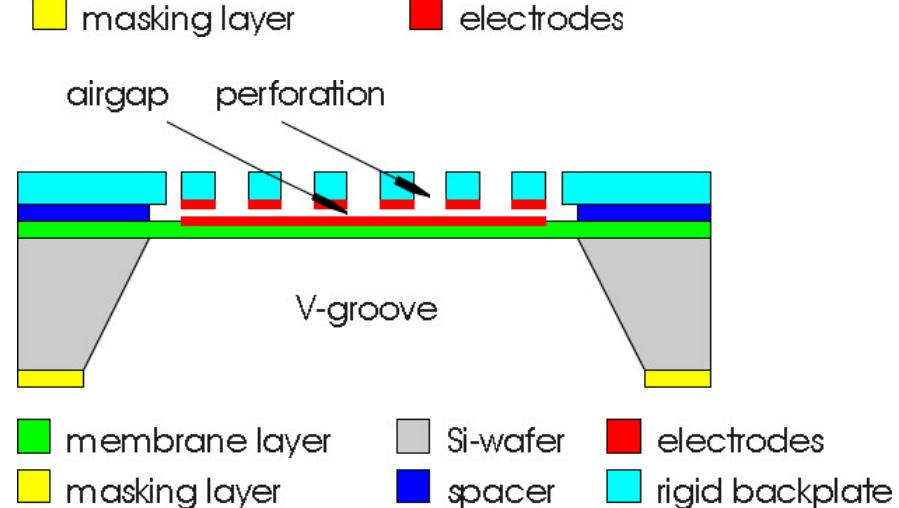
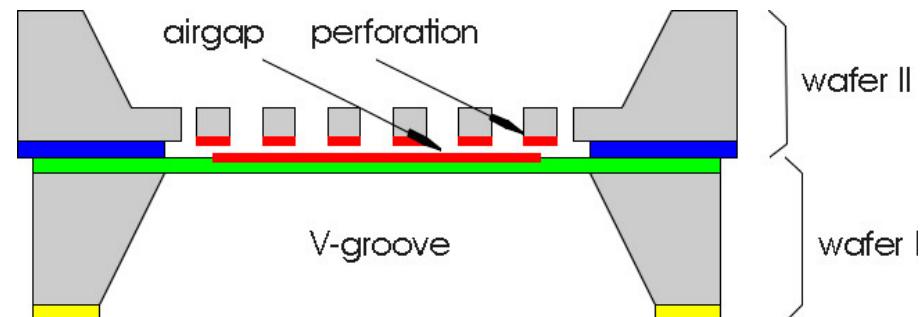
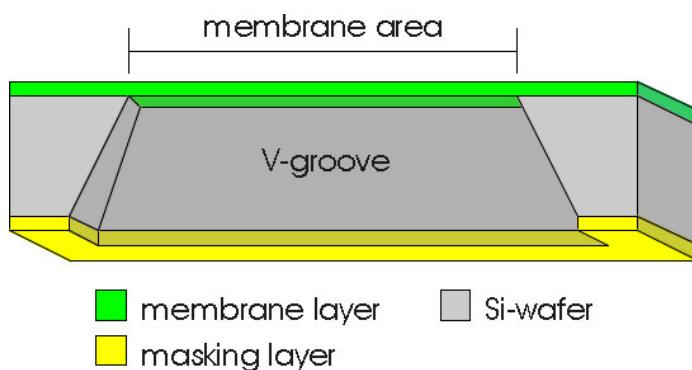
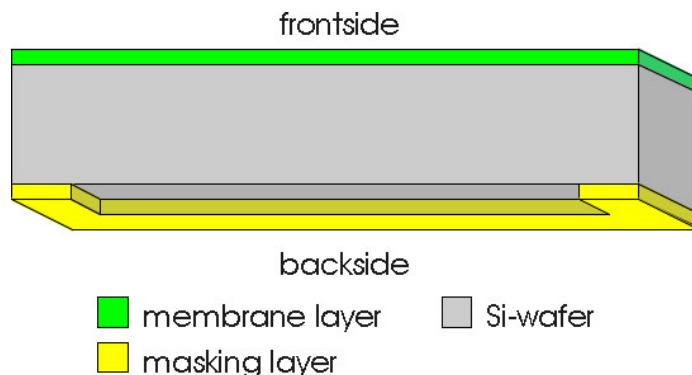
$$C_o = \epsilon \frac{A}{d}$$

$$\frac{\Delta C}{\Delta d} = -\epsilon \frac{A}{d^2}$$

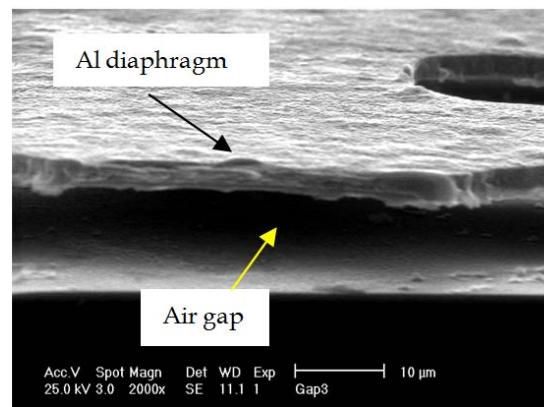
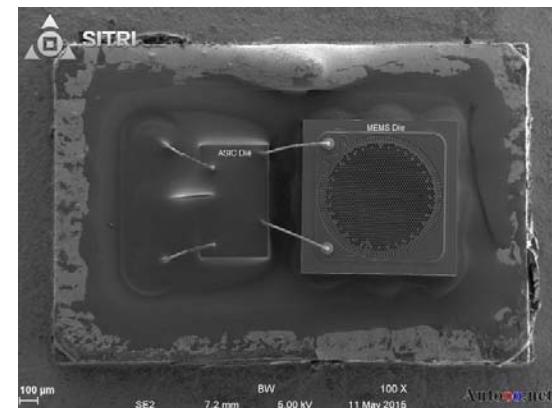
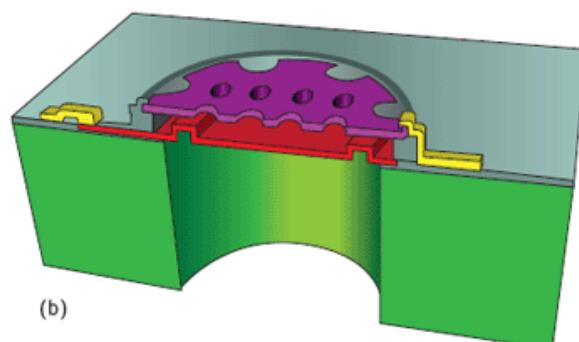
DRIE (deep reactive ion-etching)  
etched membrane



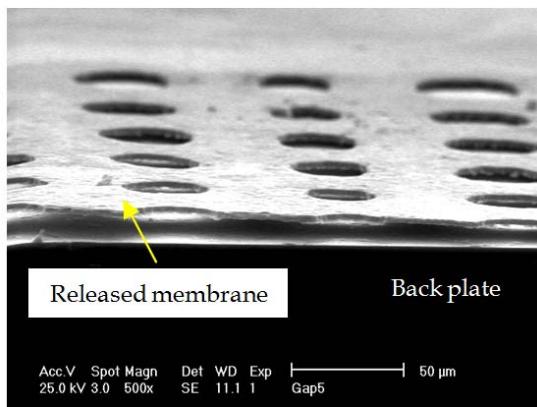
# MICROPHONE – BULK MICROMACHINING (KOH)



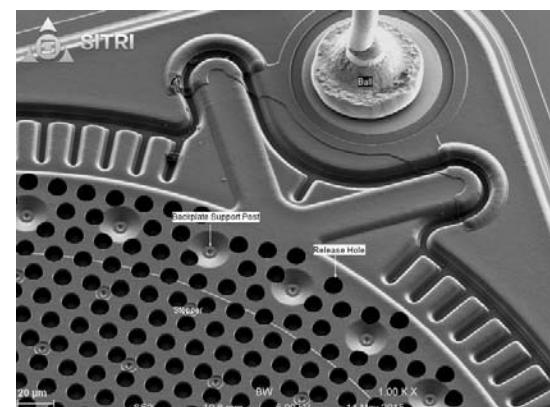
## MICROPHONE – BULK / SURFACE MICROMACHINING (COMBO)



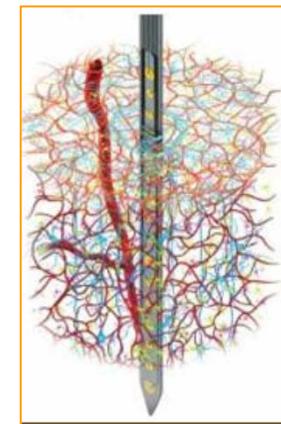
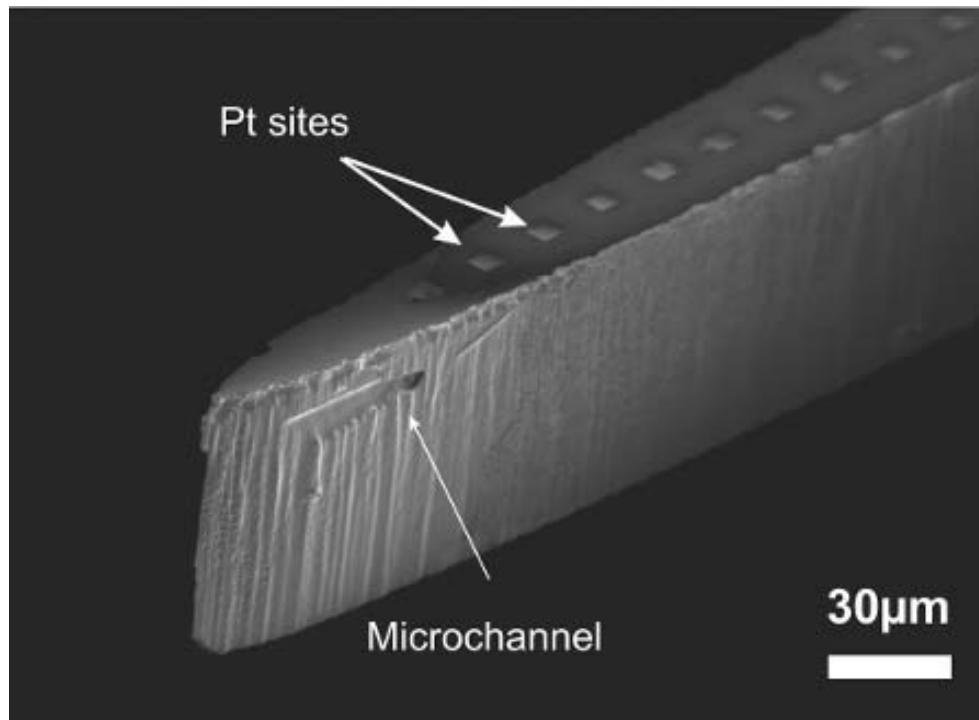
(a) Air gap of microphone



(b) Released membrane structure

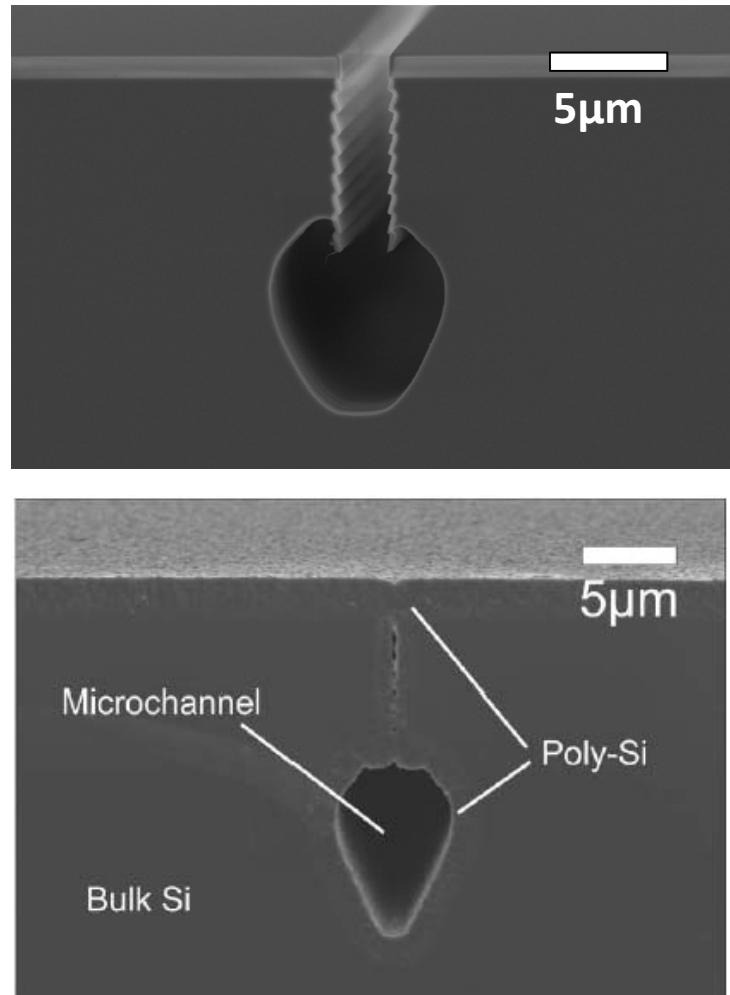
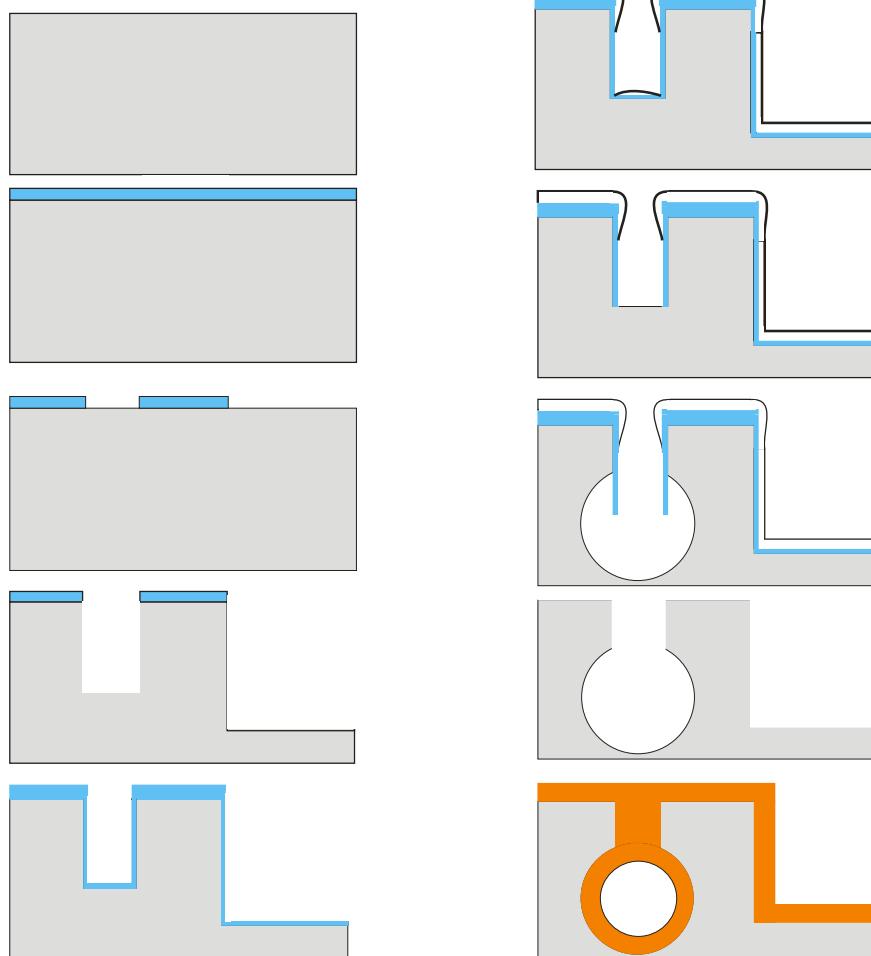


# DRUG DELIVERY CHANNELS IN SILICON NEURAL PROBE



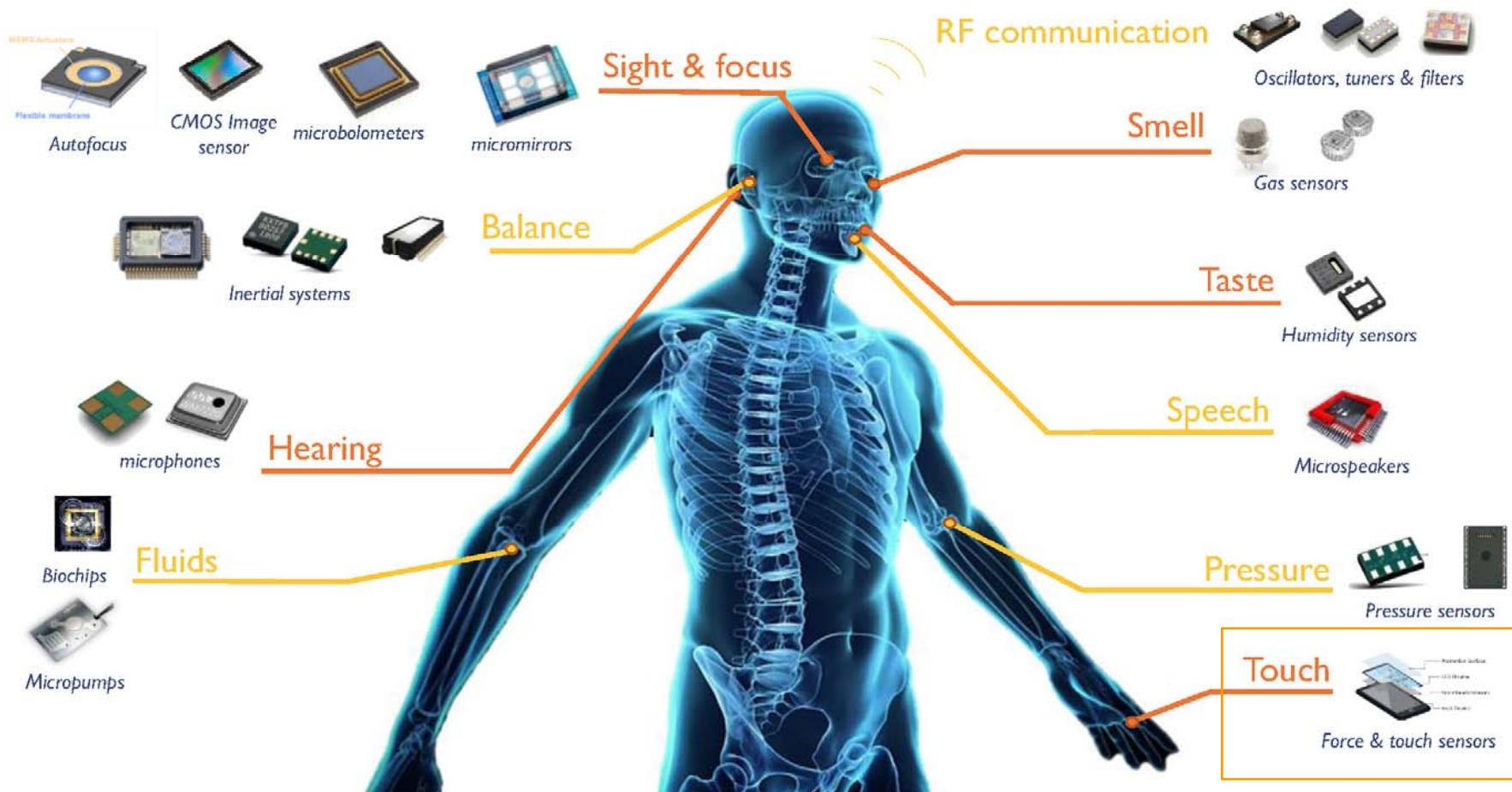
- High throughput channel array in a single substrate
  - Utilising the whole cross-section of the shaft
  - Orientation independent positioning
  - CMOS compatible fabrication technology
  - High quality surface applicable for further lithographic steps

# FABRICATION TECHNOLOGY OF BURRIED CHANNELS



# INTEGRATED MICROSYSTEMS

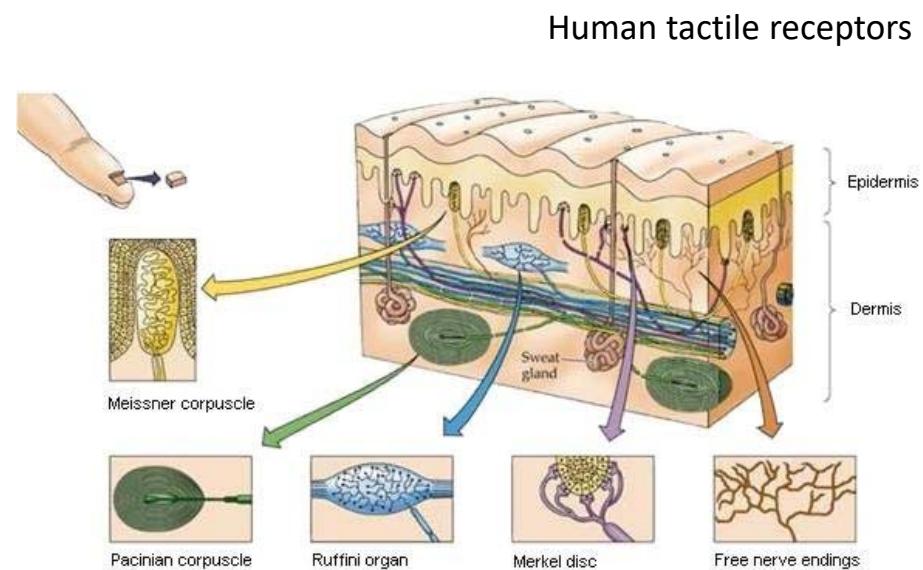
## HOW to MIMIC HUMAN SENSING?



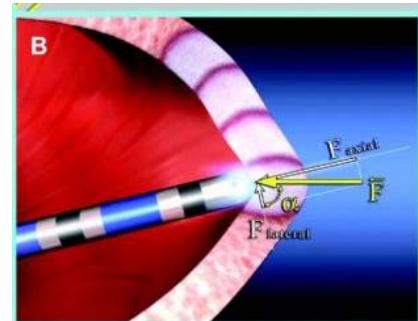
# ARTIFICIAL TACTILE RECEPTORS

## BIOMIMETIC FORCE DETECTION ANALOGY of TACTILE SENSING

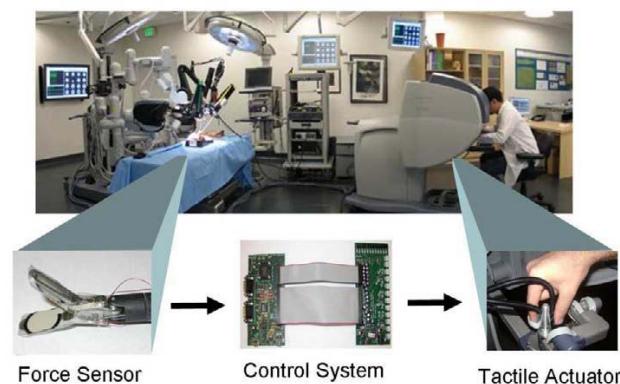
- static pressure, low and high frequency vibration, SHEAR FORCES!!!
- lubricity, roughness, patterns, shape...



## IN VIVO CONTACT FORCE MEASUREMENT



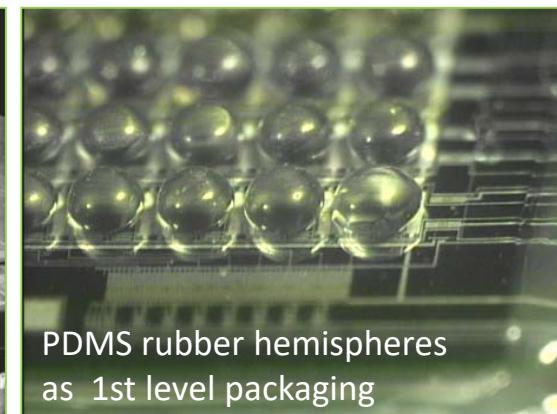
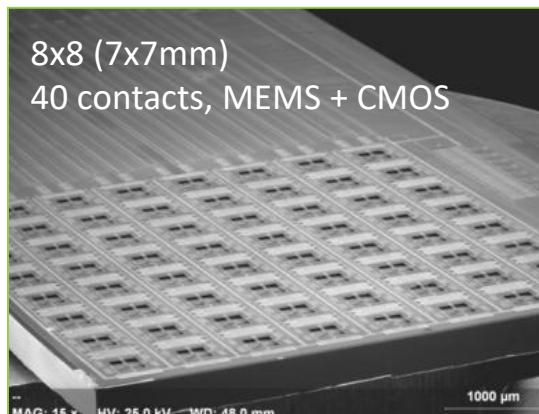
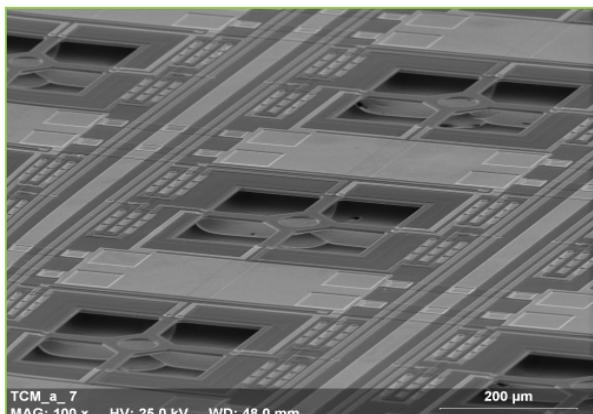
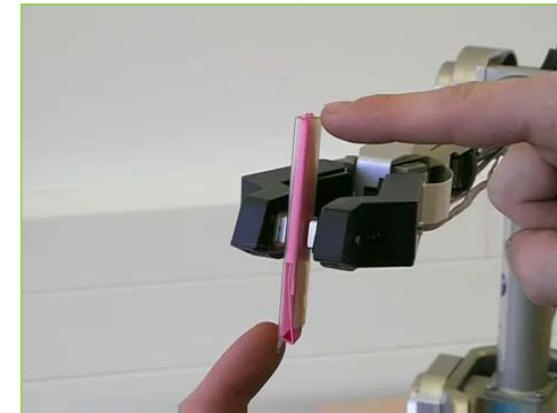
- visualization of contact force between catheter tip and the heart wall during catheter ablation
- Tactile (force) feedback during MIS surgery



## ARTIFICIAL TACTILE RECEPTORS

### MEMS BASED 3D VECTORIAL $\mu$ FORCE SENSOR

- **3D MEMS technology** based realisation of crystalline Silicon sensing elements
- **piezoresistive** read-out principle

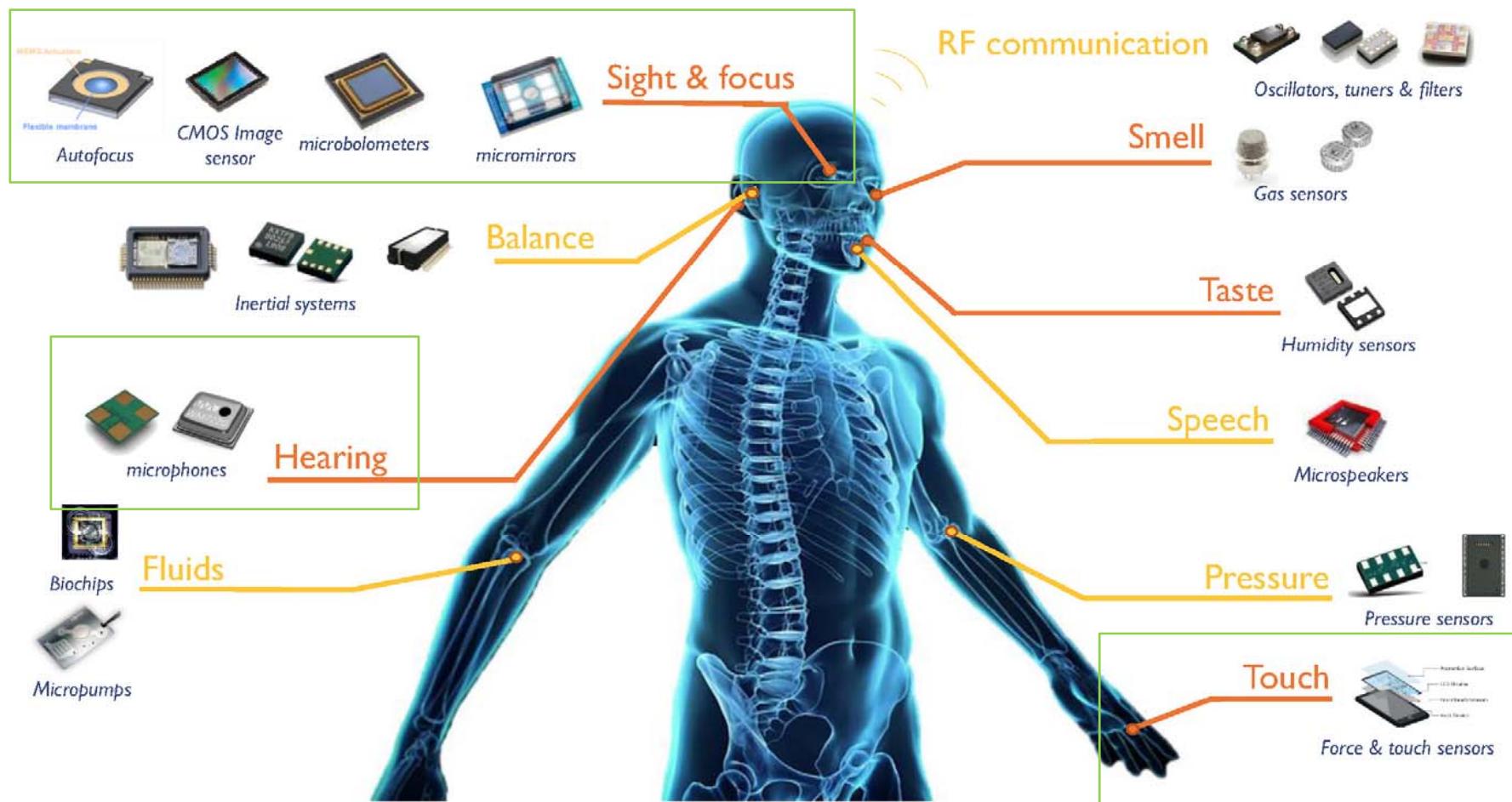


### ACTUAL FEATURES

- sensitivity and resolution: similar to human fingers
- max. density: 8x8 taxels with CMOS addressing read-out circuitry
- neuromorph flexible polymer covering

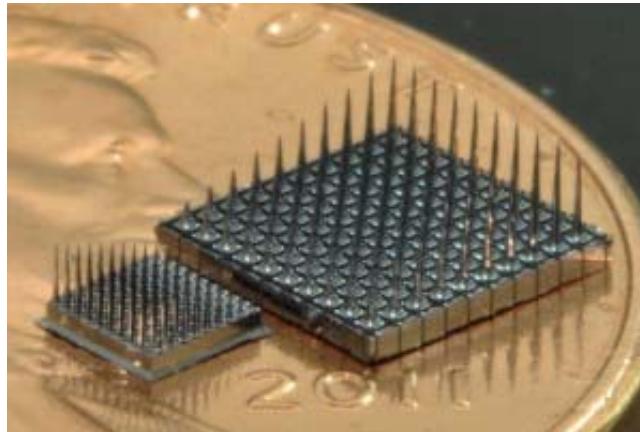
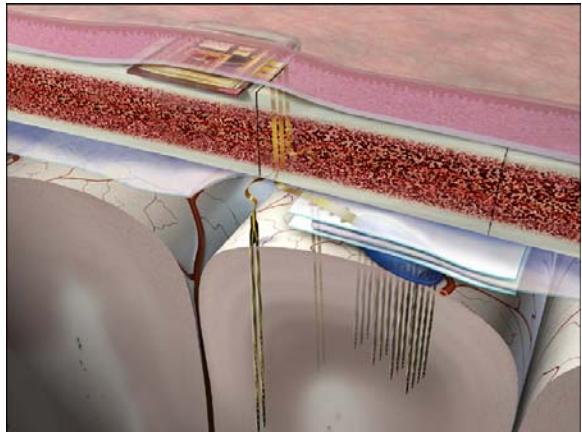


## HOW to RECOVER HUMAN SENSING OR FUNCTIONS?

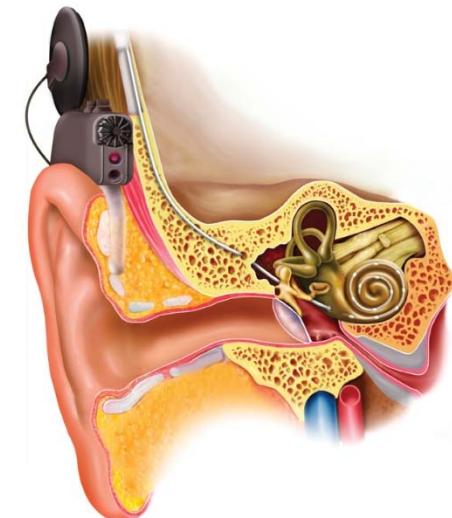


## IMPLANTABLE MICROSYSTEMS

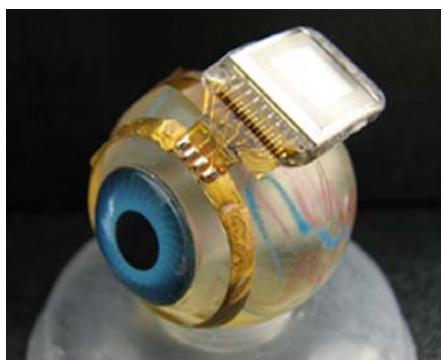
### NEURAL CELL ACTIVITY RECORDING



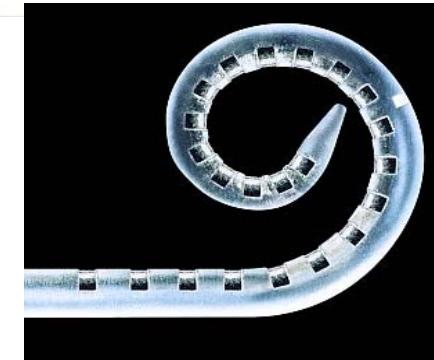
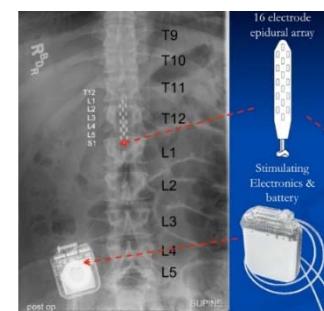
### HEARING: COCLEAR IMPLANT



### IMAGING



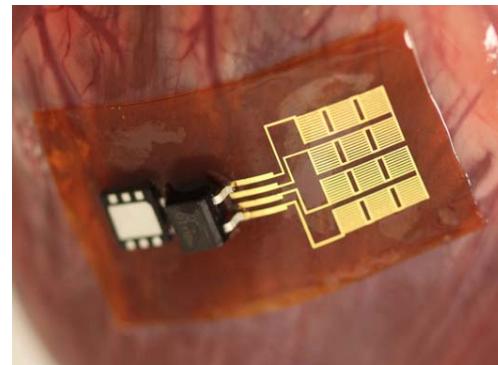
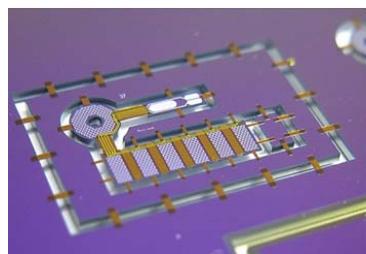
### INTERFACING / STIMULATION



## FUTURE TRENDS IN MEDICAL MEMS APPLICATIONS

### SMART SYSTEMS – HIGH DENSITY INTEGRATION    IMPLANTABLE AUTONOMOUS DEVICES

- Energy supply
- Sensing
- Signal / Data processing
- Communication
- Actuation



University of Illinois

- Energy harvesting
- SENSING: continuous health monitoring
- ACTUATION: immediate treating (drug injection)
- INTERNET of THINGS

### NANOTECHNOLOGY

- diagnostics & treating



### BRAIN-MACHINE INTERFACING



# INTRO MTA EK MFA



## MEMS laboratory:

300+150 m<sup>2</sup> clean room (4inch wafers) - 1mm resolution - mask shop (Heidelberg laser PG & direct writing),

Mask aligner / nanoimprinting system (Karl Süss MA 6, Quintel),

DRIE (Oxford Instruments Plasmalab 100),

# Physical and chemical layer deposition techniques

(vacuum evaporation, sputtering, 2x4 diffusion tubes, LPCVD,

ALD),

Wafer bonder (Karl Süss BA 6), ion implanter, etc.



## Nanoprocessing and analysis / characterisation:

E-BEAM, FIB, SEM, TEM, AFM, XPS, EDX, Auger, SIMS

## **Zeiss-SMT LEO 1540 XB SEM,**

## **Canion FIB nanoprocessing system**

## SEM and focused ion beam (FIB),

## Gas injection system (GIS) (EBAD, IBAD)

and Energy Dispersive Spectroscopy (EDS)

#### **Nanoprocessing and analysis / characterisation:**

RAITH 150 E-BEAM

- Direct writing / mask processing - Ultra high resolution
  - Thermal field emission (Schottky) source.
  - GEMINI (state-of-the-art low kV performance, beam energy: 200 V – 30 kV.
  - 6" laser interferometer stage
  - electrostatic clamping
  - automatic sample levelling by 3-points piezo motor
  - Writable surface: 0.5 – 800  $\mu\text{m}$
  - Fixed Beam Moving Stage (FBMS)
  - Fast Pattern Generator max. 10 MHz writing frequency
  - Minimum dwell time: 2 ns.
  - Measurement functions: linewidth and long-range laser interferometry based 2 nm resolution
  - Magnifications: 20 – 900.000 X.





## ACKNOWLEDGEMENT

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## SPECIAL THANKS TO



**R. E. Gyurcsányi**  
Budapest University of Technology and Economics



**T. Mészáros, P. Soós**  
Semmelweis University, Hungary



**M. Varga, B. Szabó**  
Elektronika 77 Ltd., Hungary



**J. Prechl, K. Pap**  
Eötvös Loránd University, Immunology Dept.

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FEJLESZTÉSI ÉS  
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BEFETETÉS A JÖVŐBE



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